

A STUDY OF EXHAUST GAS
TEMPERATURE MEASUREMENT METHODS

by

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M.S. Degree MIT 1947

Thesis
B38

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**A STUDY OF EXHAUST GAS
TEMPERATURE MEASUREMENT METHODS**

Submitted to the Massachusetts
Institute of Technology
May 23, 1947.

by

Lieutenant Commander Thomas M. Bennett, U.S.N.
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Lieutenant Commander Francis P. Cuccias, U.S.N.
Lieutenant Commander George C. Duncan, U.S.N.
Massachusetts Institute of Technology,
Cambridge, Massachusetts.

Dear Sir:

Submitted in Partial Fulfillment of the Requirements for the
Degree of

Master of Science in
Temperature Measurement Methods is hereby submitted

in partial fulfillment of the requirements for the

Aeronautical Engineering
Degree of Master of Science in Aeronautical Engineering.

from the

Massachusetts Institute of Technology
respectfully,

1947

Cambridge, Massachusetts,
May 23, 1947.

Rogowski, A. H. Massachusetts Institute of Technology.

Professor Joseph S. Newell,
Secretary of the Faculty,
Massachusetts Institute of Technology,
Cambridge, Massachusetts.

Liverpool, E. C.
Dear Sir:

Enc. A thesis entitled "A Study of Exhaust Gas
General Electric Company, Schenectady, N. Y.
Temperature Measurement Methods" is herewith submitted
in partial fulfillment of the requirements for the
degree of Master of Science in Aeronautical Engineering.

Cambridge, Massachusetts,
May 23, 1947.

Professor Joseph E. Howell,
Secretary of the Faculty,
Massachusetts Institute of Technology,
Cambridge, Massachusetts.

Dear Sir:

A thesis entitled "A Study of Kinetic Gas
Temperature Measurement Methods" is herewith submitted
in partial fulfillment of the requirements for the
degree of Master of Science in Mechanical Engineering.

Respectfully,

ACKNOWLEDGMENTS

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Leary, W. A. " " " "

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Ku, P. M. " " " "

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General Electric Company, Schenectady, New York.

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General Electric Company, Schenectady, New York.

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The results of these tests are shown on plots of temperature vs percent of chemically correct fuel air ratio. Final conclusions made are that the B.E. thermocouple is the most practical within the range of the instrument and that the heat exchanger method can be made more accurate than in this study and can be used over a much greater range than the thermo-

SECRET

1. The purpose of this document is to provide information regarding the status of the project. The project is currently in the planning stage and is expected to be completed by the end of the year. The project is being managed by the Project Manager, who is responsible for ensuring that the project is completed on time and within budget. The project is being funded by the Department of Defense, which is providing the necessary resources to ensure the success of the project. The project is being implemented in a phased manner, with the first phase being completed by the end of the year. The project is being monitored closely by the Project Manager, who is providing regular reports to the Department of Defense. The project is expected to have a significant impact on the Department of Defense's operations and is being considered a high priority.

SUMMARY

is not considered an ex-

An investigation was performed on a CFR, single cylinder, four stroke engine to compare three different methods of measuring the temperature of the exhaust gases. The three methods used were:

1. Computation from indicator card.
2. Direct reading of the G.E. shielded thermocouple placed at the exit of the exhaust port.
3. Computation by use of a heat exchanger.

The tests were carried out at three different compression ratios and through a range of five fuel air ratios.

The indicator card method produced the highest temperatures. The G.E. Thermocouple temperature showed an energy content of the exhaust gases lower than the energy content as computed by the indicator method by approximately 6% of the heating value of the fuel. The heat transfer method gave temperatures lower than the other two and approximately ten percent of the fuel heating value below the indicator method. Good correlation was obtained among the three methods in the temperature trends with the maximum temperature at approximately chemically correct fuel air ratio.

The results of these tests are shown as plots of temperature vs percent of chemically correct fuel air ratio. Final conclusions made are that the G.E. Thermocouple is the most practical within the range of the instrument and that the heat exchanger method can be made more accurate than in this study and can be used over a much greater range than the thermo-

An investigation was performed on a CTR, single cylinder, four stroke engine to compare three different methods of measuring the temperature of the exhaust gases. The three methods used were:

1. Direct reading of the E.K. shielded thermocouple placed at the exit of the exhaust manifold.
2. Computation by use of a heat exchanger.
3. Computation from indicator card.

The tests were carried out at three different compression ratios and through a range of five fuel air ratios. The indicator card method produced the highest temperature. The E.K. thermocouple temperature showed an energy content of the exhaust gases lower than the energy content as computed by the indicator method by approximately 6% of the heating value of the fuel. The heat transfer method gave temperatures lower than the other two and approximately ten percent of the fuel heating value below the indicator method. Good correlation was obtained among the three methods in the temperature trends with the maximum temperature at approximately chemically correct fuel air ratio.

The results of these tests are shown as plots of temperature vs percent of chemically correct fuel air ratio. Final conclusions made are that the E.K. thermocouple is the most practical within the range of the instrument and that the heat exchanger method can be made more accurate than in this study and can be used over a much greater range than the thermo-

couple. The indicator card method is not considered an actual temperature measuring device.

The measurement of energy in exhaust gases has never been successfully determined due to the inherent difficulty of measuring temperatures in a hot moving stream. Since an accurate measurement of these temperatures is necessary for engine heat balance and for the design of auxiliary turbine units, it is considered extremely important that a method be devised which will give satisfactory results.

The present investigation is a comparison between two of the most satisfactory methods to date, the General Electric shielded thermocouple (Ref. 1), and High Speed Engine Indicator card, with the heat exchanger method of this thesis.

This work was done in the Sloan Automotive Laboratory at the Massachusetts Institute of Technology in April and May, 1947.

couple. The indicator card method is not considered an

fuel temperature measuring device as a fuel temperature

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INTRODUCTION

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Air supplied to the engine was passed through a sharp-edged orifice (0.515 inch diameter - installed with flange taps according to ASME specifications -- Refs. 2 and 3), a 50 gallon surge tank, a throttle valve, a vaporizing tank, and an intake pipe leading to the engine. Temperature of the inlet mixture was controlled by supplying either low pressure steam or cooling water to the vaporizing tank jacket.

FUEL AND FUEL SYSTEM

The fuel used was one hundred octane gasoline supplied by the laboratory system. The fuel was supplied by an externally driven fuel pump and passed through a bubble separator,

INTRODUCTION

The measurement of energy in exhaust gases has never been successfully determined due to the inherent difficulty of measuring temperatures in a hot moving stream. Since an accurate measurement of these temperatures is necessary for engine heat balance and for the design of auxiliary turbine units, it is considered extremely important that a method be devised which will give satisfactory results.

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DESCRIPTION OF APPARATUS

GENERAL

The set-up and arrangement of the apparatus used in this investigation is shown in Figs. 1, 2, and 3.

ENGINE

The engine was a CFR, single cylinder, four stroke engine. It was liquid cooled, had single spark ignition, a variable compression ratio, a bore of 3.25 inches, and a stroke of 4.5 inches. It had single inlet and exhaust valves with no overlap. The engine drove, besides the dynamometer, the oil pressure pump, the ignition breaker points, the tachometer, the dynamometer exciter, and the M.I.T. high speed engine indicator.

INLET SYSTEM

Air supplied to the engine was passed through a sharp-edged orifice (0.515 inch diameter - installed with flange taps according to ASME specifications -- Refs. 2 and 3), a 50 gallon surge tank, a throttle valve, a vaporizing tank, and an intake pipe leading to the engine. Temperature of the inlet mixture was controlled by supplying either low pressure steam or cooling water to the vaporizing tank jacket.

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GENERAL

The set-up and arrangement of the apparatus used in this investigation is shown in Figs. 1, 2, and 3.

ENGINE

The engine was a GPR, single cylinder, four stroke engine. It was liquid cooled, had electric spark ignition, a variable compression ratio, a bore of 3.25 inches, and a stroke of 4.5 inches. It had single inlet and exhaust valves with no overlap. The engine drove, besides the dynamometer, the oil pressure pump, the ignition breaker points, the tachometer, the dynamometer exciter, and the M.I.T. high speed engine indicator.

INLET SYSTEM

Air supplied to the engine was passed through a sharp-edged orifice (0.515 inch diameter - installed with flange taps according to ASME specifications -- Refs. 2 and 3), a 50 gallon surge tank, a throttle valve, a vaporizing tank, and an intake pipe leading to the engine. Temperature of the inlet mixture was controlled by supplying either low pressure steam or cooling water to the vaporizing tank jacket.

FUEL AND FUEL SYSTEM

The fuel used was one hundred octane gasoline supplied by the laboratory system. The fuel was supplied by an externally driven fuel pump and passed through a bubble separator,

a rotometer, and a needle valve directly over the vaporizing tank. This valve was used to pressurize the supply line to avoid vapor lock. Vaporization was obtained by heating the vaporizing tank with steam (Fig. 4).

CYLINDER COOLING SYSTEM

Cooling was provided by circulating water through the engine. Air flow was measured with a standard water jacket by an externally driven circulating pump. The pump-cooled orifice constructed in accordance with the ASME temperature was maintained constant at boiling point and Field Meter Committee specifications. A differential water excessive boiling was prevented by cooling coils.

A large electric orifice-dynamometer was used to measure

LUBRICATING SYSTEM

The oil pressure was obtained through an engine-driven pump. The oil circulating pump was externally driven. The oil temperature was satisfactorily controlled by a heat exchanger using low pressure steam or water. This rotometer

are shown in Fig. 9.

IGNITION SYSTEM

Ignition was provided by a breaker operating at one-half crankshaft speed. The electrical circuit is shown in Fig. 5.

Spark advance was measured by a neon light (excited by

EXHAUST SYSTEM

The exhaust gases were discharged from the engine into a 1-1/4 inch iron pipe expanding into a 2 inch pipe in which the General Electric shielded thermocouple was housed (Fig. 6). From here the gases were passed through a flexible steel coupling into a heat exchanger for cooling purposes. The gases from the heat exchanger discharged into a surge tank and thence exhausted to the trench. The entire system from

a rotometer, and a needle valve directly over the vaporizing tank. This valve was used to pressurize the supply line to avoid vapor lock. Vaporization was obtained by heating the vaporizing tank with steam (Fig. 4).

CYLINDER COOLING SYSTEM

Cooling was provided by circulating water through the water jacket by an externally driven circulating pump. The temperature was maintained constant at boiling point and excessive boiling was prevented by cooling coils.

LUBRICATING SYSTEM

The oil pressure was obtained through an engine-driven pump. The oil circulating pump was externally driven. The oil temperature was automatically controlled by a heat exchanger using low pressure steam or water.

IGNITION SYSTEM

Ignition was provided by a breaker operating at one-half crankshaft speed. The electrical circuit is shown in Fig. 5.

EXHAUST SYSTEM

The exhaust gases were discharged from the engine into a 1-1/4 inch iron pipe expanding into a 3 inch pipe in which the General Electric shielded thermocouple was housed (Fig. 6). From here the gases were passed through a flexible steel coupling into a heat exchanger for cooling purposes. The gases from the heat exchanger discharged into a surge tank and thence exhausted to the trench. The entire system from

cylinder to discharge was lagged to reduce heat loss to a minimum. A detailed sketch of the exhaust system is shown in Fig. 3.

measured by an iron-constantan thermocouple in the
exhaust. Fuel temperature was measured directly after

MEASURING INSTRUMENTS

A large electric cradle-dynamometer was used to measure torque (Fig. 7). Speed was determined by a calibrated mechanical tachometer. Air flow was measured with a standard sharp-edged orifice constructed in accordance with the ASME Fluid Meter Committee specifications. A differential water manometer was used to measure the pressure difference across the orifice. The orifice diameter was 0.515 inches with standard flange pressure taps. The calibration and corrections for temperature and pressure for this set-up are shown in Fig. 8. Fuel flow was measured with a Fischer and Porter Stabl-Vis Rotometer; calibration curves for this rotometer are shown in Fig. 9.

Atmospheric pressure was measured by a mercury barometer; inlet pressure was measured in the vaporizing tank by a water manometer, as was the exhaust pressure in the exhaust surge tank. Spark advance was measured by a neon light (excited by the spark discharge) which revolved with the crankshaft.

Cylinder inlet temperature was measured by a mercury bulb thermometer in the inlet pipe. Circulating water temperatures through the exhaust gas heat exchanger were measured by mercury bulb thermometers at intake and outlet. Exhaust gas temperatures were measured in four places; one, shortly after the exhaust port of the engine by the General Electric

cylinder to discharge was lagged to reduce heat loss to a minimum. A detailed sketch of the exhaust system is shown in Fig. 3.

MEASURING INSTRUMENTS

A large electric cradle-dynamometer was used to measure torque (Fig. 7). Speed was determined by a calibrated mechanical tachometer. Air flow was measured with a standard sharp-edged orifice connected in accordance with the ASME Fluid Meter Committee specifications. A differential water manometer was used to measure the pressure difference across the orifice. The orifice diameter was 0.25 inches with standard flange pressure taps. The calibration and corrections for temperature and pressure for this set-up are shown in Fig. 8. Fuel flow was measured with a Viscometer and Porter-Stahl-Via Rotometer; calibration curves for this rotometer

are shown in Fig. 9.

Atmospheric pressure was measured by a mercury barometer; inlet pressure was measured in the vaporizing tank by a water manometer, as was the exhaust pressure in the exhaust surge tank. Spark advance was measured by a neon light (excited by the spark discharge) which revolved with the crankshaft.

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shielded thermocouple, and at three places in the exhaust

surge tank by iron-constantan thermocouples. Oil tempera-

The primary consideration of this investigation was to correlate the results obtained from the three methods used. Fuel temperature was measured directly after the rotometer by a mercury bulb thermometer. The M.I.T. High Speed Indicator was used to obtain records of cylinder pressure versus crank angles. (Fig. 10). The M.I.T. transfer table was used to obtain pressure-volume diagrams therefrom. Compression ratios with five different fuel-air ratios for each compression ratio. Compression ratios used were 7, 8, and 9, and fuel-air ratios used were 60%, 90%, 100%, 110%, and 120% of chemically correct fuel-air ratios.

The following conditions were held at constant values as indicated:

1. Oil pressure	40 $\frac{1}{2}$ psia.
2. Jacket water temperature	212°F
3. Inlet valve clearance	.012"
4. Exhaust valve clearance	.014"
5. Engine speed	1500 \pm 10 rpm
6. Spark advance	20°

The following factors were varied as necessary for test:

1. Fuel-air ratio
2. Compression ratio

Many familiarization runs were made to determine the best operating conditions. During these familiarization runs many difficulties were discovered which had to be overcome in order to proceed. First, the heat exchanger, used to cool the exhaust gases, was much more efficient than originally

which thermocouple, and at three places in the exhaust
exhaust tank by iron-constantan thermocouples. Oil tempera-
ture was measured by an iron-constantan thermocouple in the
crankcase. Fuel temperature was measured directly after
the rotometer by a mercury bulb thermometer. The W.I.T.
High Speed Indicator was used to obtain records of cylinder
pressure versus crank angles (Fig. 10). The W.I.T. transfer
table was used to obtain pressure-volume diagrams therefrom.
Third order Fourier series calculations were made of the
records obtained and the results are shown in Figs. 11 and 12.
The cylinder wall cooling was measured by a thermocouple
located in the cylinder wall. The thermocouple was connected
to a potentiometer and the output was connected to a recorder
in the W.I.T. Fuel flow was measured with a Venturi flow
meter. The mechanical efficiency was measured by a torque
meter and shown in Fig. 13.
A complete description of the apparatus and the results of the
tests are given in the report. The results of the tests are
summarized in the following table. The first column gives the
test conditions and the second column gives the results of the
tests. The third column gives the results of the calculations.
The data are given in the following table.

PROCEDURE

The primary consideration of this investigation was to correlate the results obtained from the three methods used to measure exhaust gas temperatures. This was done by taking simultaneous readings of all the data necessary, as shown in Table I. The runs made were divided into groups of three different compression ratios with five different fuel-air ratios for each compression ratio. Compression ratios used were 7, 8, and 9, and fuel-air ratios used were 80%, 90%, 100%, 110%, and 120% of chemically correct fuel-air ratios.

The following conditions were held at constant values as indicated:

1. Oil pressure 40 ± 2 psia.
2. Jacket water temperature 212°F
3. Inlet valve clearance $.012''$
4. Exhaust valve clearance $.014''$
5. Engine speed 1600 ± 10 rpm
6. Spark advance 20°

The following factors were varied as necessary for test:

1. Fuel-air ratio
2. Compression ratio

Many familiarization runs were made to determine the best operating conditions. During these familiarization runs many difficulties were discovered which had to be overcome in order to proceed. First, the heat exchanger, used to cool the exhaust gases, was much more efficient than originally

PROCEDURE

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as indicated:

1. Oil pressure 40-45 psi.
2. Jacket water temperature 212°F
3. Inlet valve clearance .015"
4. Exhaust valve clearance .014"
5. Engine speed 1600-1800 rpm
6. Spark advance 20°

The following factors were varied as necessary for test:

1. Fuel-air ratio
2. Compression ratio

Many familiarization runs were made to determine the best operating conditions. During these familiarization runs many difficulties were discovered which had to be overcome in order to proceed. First, the heat exchanger, used to cool the exhaust gases, was much more efficient than originally

designed (according to Ref. 4), and the cooler had to be decreased in size from 12' to 8', and from counter flow to parallel flow, to maintain a temperature well above 212° F. in the exhaust surge tank. Also, due to excessive heat caused by lagging the exhaust pipes, a sodium-filled exhaust valve was used to maintain cooler valve temperature to lessen the probability of backfire. The third method was the direct temperature reading of the General Electric thermocouple. A description of the use of the Motter charts and the high speed indicator diagram for the solution of this problem is given below.

To compute the exhaust gas temperature from a high speed indicator card the First Law of Thermodynamics was used. From a point "X" on the expansion stroke line immediately before the exhaust valve opens an equivalent cycle was used through the blowdown and exhaust processes.

It was necessary to find a relation between the cylinder volume and the Motter Chart volume, and to determine the percentage of residual gas "r" in the actual cycle. These two must be done together by trial and error.

Relation between V_{chart} and V_{cylinder}

In actual cycle:

Fresh air/cycle = 3 pounds

Air in residual gas/cycle = 1 pound

Total air/cycle = (3 + 1) pounds

Total fuel/cycle = (3 + 1)/F pounds, where F = fuel-air ratio

Total charge/cycle = (3 + 1)(1 + F) pounds

Total residual gas/cycle = 2(1 + F) pounds

designed (according to Ref. 4), and the cooler had to be de-
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 valve was used to maintain cooler valve temperature to lessen
 the probability of backfire.

The following conditions were met by exhaust valves
 used in the engine test cell:

1. All valves
2. Exhaust valve temperature
3. Exhaust valve diameter
4. Exhaust valve material
5. Exhaust valve design
6. Exhaust valve location
7. Exhaust valve operation
8. Exhaust valve maintenance

The following conditions were met by exhaust valves
 used in the engine test cell:

1. Exhaust valve diameter
2. Exhaust valve material
3. Exhaust valve design
4. Exhaust valve location
5. Exhaust valve operation
6. Exhaust valve maintenance
7. Exhaust valve temperature
8. Exhaust valve lagging

DISCUSSION

In this thesis three methods of exhaust gas temperature measurements were investigated. Two methods, namely, determination of temperature from the indicator diagram and from the heat exchanger required computation and use of the Hottel Burned Mixture Charts for (CH_2) (Fig. 11). The third method was the direct temperature reading of the General Electric Thermocouple. A description of the use of the Hottel charts and the high speed indicator diagram for the solution of this problem is given below.

To compute the exhaust gas temperature from a high speed indicator card the First Law of Thermodynamics was used. From a point "X" on the expansion stroke line immediately before the exhaust valve opens an equivalent cycle was used through the blowdown and exhaust processes.

It was necessary to find a relation between the cylinder volume and the Hottel Chart volume, and to determine the percentage of residual gas "r" in the actual cycle. These two must be done together by trial and error.

Relation between V_{chart} and $V_{cylinder}$.

In actual cycle:

Fresh air/cycle = B pounds

Air in residual gas/cycle = Z pounds

Total air/cycle = $(B + Z)$ pounds

Total fuel/cycle = $(B + Z)F$ pounds, where F = fuel-air ratio

Total charge/cycle = $(B + Z)(1 + F)$ pounds

Total residual gas/cycle = $Z(1 + F)$ pounds

DISCUSSION

In this thesis three methods of exhaust gas temperature measurements were investigated. Two methods, namely, determination of temperature from the indicator diagram and from the heat exchanger required computation and use of the Hottel burned mixture charts for (CH_2) (Fig. 11). The third method was the direct temperature reading of the General Electric Thermocouple. A description of the use of the Hottel charts and the high speed indicator diagram for the solution of this problem is given below.

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It was necessary to find a relation between the cylinder volume and the Hottel Chart volume, and to determine the percentage of residual gas "r" in the actual cycle. These two must be done together by trial and error.

Relation between V_{chart} and V_{cylinder}

In actual cycle:

$$\text{Fresh air/cycle} = B \text{ pounds}$$

$$\text{Air in residual gas/cycle} = Z \text{ pounds}$$

$$\text{Total air/cycle} = (B + Z) \text{ pounds}$$

$$\text{Total fuel/cycle} = (B + Z)F \text{ pounds, where } F = \text{fuel-air ratio}$$

$$\text{Total charge/cycle} = (B + Z)(1 + F) \text{ pounds}$$

$$\text{Total residual gas/cycle} = Z(1 + F) \text{ pounds}$$

$$\text{But } f = \frac{Z(1+F)}{(B+Z)(1+F)} = \frac{Z}{B+Z}$$

So that $Z = \frac{Bf}{1-f}$ where f is the fraction of the exhaust gas determined from the total chart.

$$\text{Total charge/cycle} = (B+Z)(1+F) = (B + \frac{Bf}{1-f})(1+F) = \frac{B}{1-f}(1+F)$$

Chart mass = $(1+F)$ pounds

$$\text{Therefore } \frac{\text{chart mass}}{\text{cycle mass}} = \frac{1-f(1+F)}{B(1+F)} = \frac{1-f}{B}$$

And $\frac{V_{\text{chart}}}{V_{\text{cylinder}}} = \frac{1-f}{B}$ Since the basis of the Hottel chart is 1 lb of gas at 14.7 psia and 520°R.

Thus a value for f was assumed and the V_{cylinder} for point "X" converted to V_{chart} . This value and the pressure at point "X" were then used to locate the corresponding point on the Hottel Chart for the same fuel-air ratio.

Then expand adiabatically to the exhaust pressure to get the volume of the burned charge at exhaust pressure on the

Now let $\frac{V_{\text{clearance}}}{V_{\text{exhaust}}} = \text{entropy of } 0.30$. This entropy was used

because the actual point was off the chart. This intro- This value of f was then compared with the assumed value, and since no error because in this region entropy is a function of temperature only. By adding the heat extracted equal.

(Btu/lb air) the enthalpy of the gases at exhaust was found.

Now the point "X" could be accurately located on the Hottel chart. By expanding to total cylinder volume converted

to chart volume, and to exhaust pressure, two points "4" and "5" respectively, could be located on the Hottel chart.

Assuming that the chemical energy of the exhaust gas did not change during the exhaust process, the sensible enthalpy of the exhaust gas may be calculated as follows: to prevent

$$\text{But } t = \frac{(1 + \frac{B}{1-t})}{(1 + \frac{B}{1-t})} = \frac{1}{1 + \frac{B}{1-t}}$$

$$\text{So that } t = \frac{B}{1-t}$$

$$\text{Total charge/cycle} = (1 + \frac{B}{1-t}) = (1 + \frac{B}{1-t}) = \frac{B}{1-t} (1 + \frac{B}{1-t})$$

$$\text{Chart mass} = (1 + \frac{B}{1-t}) \text{ pounds}$$

$$\text{Therefore } \frac{\text{chart mass}}{\text{cycle mass}} = \frac{1-t(1+\frac{B}{1-t})}{1-t} = \frac{1-t}{1-t}$$

$$\text{And } \frac{V_{\text{chart}}}{V_{\text{cylinder}}} = \frac{1-t}{1-t}$$

Thus a value for t was assumed and the V_{cylinder} for point "X" converted to V_{chart} . This value and the pressure at point "X" were then used to locate the corresponding point on the Hotell Chart for the same fuel-air ratio. Then expand adiabatically to the exhaust pressure to get the volume of the burned charge at exhaust pressure.

$$\text{Now } t = \frac{V_{\text{exhaust}}}{V_{\text{chart}}}$$

This value of t was then compared with the assumed value, and the method of determining t repeated until the two values were equal.

Now the point "X" could be accurately located on the Hotell chart. By expanding to total cylinder volume converted to chart volume, and to exhaust pressure, two points "4" and "5" respectively, could be located on the Hotell chart.

Assuming that the chemical energy of the exhaust gas did not change during the exhaust process, the sensible enthalpy of the exhaust gas may be calculated as follows:

$$(1-f)H_{se} = E_{s4} - fE_{s5} + \frac{P_e(V_1 - V_2)}{J}$$

and the temperature of the exhaust gas determined from the Hottel chart.

Determination of Exhaust Gas Temperature by the Heat Extraction Method

The heat extracted from exhaust gases (Btu/sec) by the water was found from the rate of water flow (lbs/sec) and its rise in temperature. Since the basis of the Hottel charts is one pound of air plus the necessary fuel to give the proper fuel air ratio, the Btu's/sec found from the water flow was converted to Btu's/lb air by dividing this value by the mass rate flow of air in lbs/sec.

Using the mean temperature of the gases in the surge tank, the enthalpy of these gases was determined from the Hottel chart at an entropy of 0.30. This entropy was used because the actual point was off the chart. This introduced no error because in this region enthalpy is a function of temperature only. By adding the heat extracted (Btu/lb air) the enthalpy of the gases at exhaust was found. From this value the temperature was read directly.

Design Considerations

The design of the heat exchanger for cooling was critical. The area had to be such that a reasonable rise in water temperature was produced (to minimize errors in reading); that the rate of water flow was sufficient to prevent

$$\frac{p_2(V_1 - V_2)}{2} + E_{24} - E_{23} = E_{21} \quad (1-1)$$

and the temperature of the exhaust gas determined from the

Hotel chart.

Determination of Exhaust Gas Temperature by the Heat Extraction Method

The heat extracted from exhaust gases (Btu/sec) by the water was found from the rate of water flow (lba/sec) and its rise in temperature. Since the basis of the Hotel chart is one pound of air plus the necessary fuel to give the proper fuel air ratio, the Btu/sec found from the water flow was converted to Btu/lb air by dividing this value by the mass rate flow of air in lba/sec.

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Design Considerations

The design of the heat exchanger for cooling was critical. The area had to be such that a reasonable rise in water temperature was produced (to minimize errors in reading); that the rate of water flow was sufficient to prevent

steam pockets; and that the temperature of the exhaust gases in the surge tank was sufficiently high to avoid heat water condensation.

Three ninety degree turns were made in the exhaust system before the heat exchanger. This was deemed necessary in order to prevent the shielded thermocouple from "seeing" the hot exhaust port or the entrance to the brine cooler. Also, these turns induced turbulence and helped to prevent stratification of the exhaust gases. To minimize the possibility of a large temperature gradient across the thermocouple, a third turn was introduced immediately before the cooler. This turn directed the exhaust gases vertically upward into the cooler which allowed easy removal of air from the cooling water system.

The heat exchanger consisted of a two-inch pipe concentrically placed in a three-inch pipe. One bend was made in the cooler making the discharge horizontal for convenience of installation and instrumentation. This was done by spider spacing around the inner pipe, tightly packing sand between the pipes, and cold bending the combination. This method proved very satisfactory and is recommended for any similar construction.

The surge tank was used to bring the exhaust gases to a sufficiently stagnant condition to allow accurate pressure and temperature measurements.

The whole system was lagged with high temperature insulation and it very effectively minimized heat losses to the

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mediately before the cooler. This turn directed the ex-

haust gases vertically upward into the cooler which allowed

easy removal of air from the cooling water system.

The heat exchanger consisted of a two-inch pipe

concentrically placed in a three-inch pipe. One end was

made in the cooler making the discharge horizontal for

convenience of installation and instrumentation. This was

done by splicing around the inner pipe, tightly pack-

ing sand between the pipes, and cold bonding the combination.

This method proved very satisfactory and is recommended for

any similar construction.

The surge tank was used to bring the exhaust gases to

a sufficiently stagnant condition to allow accurate pressure

and temperature measurements.

The whole system was lagged with high temperature insu-

lation and it very effectively minimized heat losses to the

atmosphere. Due to this lagging, it is recommended that the exhaust pipes and flexible coupling prior to the heat exchanger be made of high temperature steel, if the use of a measuring thermocouple is anticipated. The use of a flexible coupling between the exhaust pipe and heat exchanger was considered advisable in this investigation to prevent damage to the engine cylinder head due to vibration between the engine mount and the heavy cooling system, and to allow for changing of the compression ratio without moving the heat exchanger.

can not used for each compression ratio and fuel air ratio.

The results from the heat exchanger method used in this investigation were not quite in agreement with the direct reading of the G.E. thermocouple and were well out of agreement with the indicator method. This was expected due to the simplifying assumptions made, namely, that the exhaust system process was adiabatic and the "blow-down" was isentropic.

The temperatures computed by the indicator method were fundamentally theoretical and are not to be considered the actual temperatures existing in the exhaust pipe. However, the nearer to this temperature that any method can come, the better the method. From the curves, the temperature difference between the theoretical and the computed temperature is on the order of 400° F. This is not considered to be close enough to give an accurate heat balance of the engine, nor to predict accurately the

atmosphere. Due to this lagging, it is recommended that the exhaust pipes and flexible coupling prior to the heat exchanger be made of high temperature steel, if the use of a measuring thermocouple is anticipated. The use of a flexible coupling between the exhaust pipe and heat exchanger was considered advisable in this investigation to prevent damage to the engine cylinder head due to vibration between the engine mount and the heavy cooling system, and to allow for changing of the compression ratio without moving the heat exchanger.

The heat exchanger was of the type known as a "shell and tube" type. It consisted of a cylindrical shell with a bundle of tubes inside. The tubes were of the same diameter as the shell, and the shell was of a larger diameter. The tubes were connected to the engine cylinder head and the heat exchanger. The shell was connected to the heat exchanger. The tubes were of the same diameter as the shell, and the shell was of a larger diameter. The tubes were connected to the engine cylinder head and the heat exchanger. The shell was connected to the heat exchanger.

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RESULTS

The temperatures were computed according to the methods described in the Discussion and were plotted in Figs. 12-17. As can be seen from the curves, the trend of temperature change by each method agreed exceedingly well. Since a spark advance of 20 degrees was used in all runs, the shape of the curves for each set of compression ratios differed from one another. This was apparently due to the fact that best power spark advance was not used for each compression ratio and fuel air ratio.

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The temperatures computed by the indicator method were fundamentally theoretical and are not to be considered the actual temperatures existing in the exhaust pipe. However, the nearer to this temperature that any method can come, the better the method. From the curves, the temperature difference between the theoretical and the computed temperature is on the order of 400°F . This is not considered to be close enough to give an accurate heat balance of the engine, nor to predict accurately the

energy remaining in the exhaust gases. No accurate heat balance has ever been performed on an internal combustion engine, but according to theoretical considerations, approximately 15% to 20% of the heating value of the fuel is dissipated in the cylinder and jacket. Because the exhaust pipe and port in this investigation were at such a high temperature, due to the lagging, it was expected that the heat transfer back to the cylinder from the exhaust pipe would appreciably increase the amount indicated above. Therefore, by taking an estimated 8% to 11% of the heating value of the fuel, additionally dissipated to the cylinder and jacket, and applying this correction to the curves already obtained by the cooler method, they can be made to fall directly on the indicator curves.

The G.E. thermocouple gave results that were closer to the indicator temperatures, the average error from this temperature being about 6% to 7% of the heating value of the fuel. Because of its small size and simple installation, this thermocouple was considered both practical and accurate. Therefore, it is concluded that this method is, to date, the most satisfactory method of measuring high temperatures in a hot moving stream within the limits of the instrument, that is, up to 1800 degrees Fahrenheit.

Since the error between the G.E. thermocouple reading and the heat exchanger method is only about three percent of the heating value of the fuel, and since this error can

energy remaining in the exhaust gases. No accurate heat balance has ever been performed on an internal combustion engine, but according to theoretical considerations, approximately 15% to 20% of the heating value of the fuel is dissipated in the cylinder and jacket. Because the exhaust pipe and port in this investigation were at such a high temperature, due to the lagging, it was expected that the heat transfer back to the cylinder from the exhaust pipe would appreciably increase the amount indicated above. Therefore, by taking an estimated 2% to 1% of the heating value of the fuel, additionally dissipated to the cylinder and jacket, and applying this correction to the curves already obtained by the cooler method, they can be made to fall directly on the indicator curves. The G.E. thermocouple gave results that were closer to the indicator temperatures, the average error from this temperature being about 6% to 7% of the heating value of the fuel. Because of its small size and simple installation, this thermocouple was considered both practical and accurate. Therefore, it is concluded that this method is, to date, the most satisfactory method of measuring high temperatures in a hot moving stream within the limits of the instrument, that is, up to 1300 degrees Fahrenheit. Since the error between the G.E. thermocouple reading and the heat exchanger method is only about three percent of the heating value of the fuel, and since this error can

be further reduced or eliminated by connecting the heat exchanger, with an insulating coupling, directly to the exhaust port of the engine, the method developed in this thesis is recommended for use with engines whose exhaust temperatures are both in and above the temperature range of the G.E. thermocouple. This is based on the assumption that the G.E. thermocouple gave results closest to the actual conditions.

3. The heat exchanger method is sufficiently accurate for the measurement of temperatures above the range of the G.E. thermocouple.
4. The indicator method can be used for rough results by subtracting approximately 4% of the heating value of the fuel from the energy of the exhaust gases determined by this method.

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CONCLUSIONS

1. Satisfactory correlation was obtained among the three methods investigated.
2. The General Electric shielded thermocouple gave values closest to the expected results and is the most practical reliable method.
3. The heat exchanger method is sufficiently accurate for the measurement of temperatures above the range of the G.E. thermocouple.
4. The indicator method can be used for rough results by subtracting approximately 6% of the heating value of the fuel from the energy of the exhaust gases determined by this method.

CONCLUSIONS

1. Satisfactory correlation was obtained among the three methods investigated.
2. The General Electric shielded thermocouple gave values closest to the expected results and is the most practical reliable method.
3. The heat exchanger method is sufficiently accurate for the measurement of temperatures above the range of the G.E. thermocouple.
4. The indicator method can be used for rough results by subtracting approximately 6% of the heating value of the fuel from the energy of the exhaust gases determined by this method.

RECOMMENDATIONS

1. This investigation should be further carried on, connecting the water cooler (by insulated coupling), directly to the exhaust port to further reduce heat losses.
2. "Field Meters, Their Theory and Application", Part 1.
2. To establish a definite trend of the temperature curves, best power spark advance should be used for all runs.
3. "ASME Power Test Codes of 1940", Part 5, Chapter 4.
3. If a similar set-up is used, high temperature metals should be used for the portion of the exhaust pipe between engine and cooler.
4. "McGraw-Hill Test Company, 1942.
5. Taylor, G. F., and Taylor, R. S.: "The Internal Combustion Engine", International Textbook Company, Scranton, Pa., 1938.
6. "Temperature - Its Measurement and Control in Science and Industry", Reinhold Publishing Corp., 1941.

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REFERENCES

1. King, W. J.: "Measurement of High Temperatures in High Velocity Gas Streams", Transactions, ASME, Vol. 65, 1943, pp. 421-431.
 Heat temperature in exhaust gases tank (°F) 100
 Mass rate of water flow (lb/min) 3.42
2. "Fluid Meters, Their Theory and Application", Part I, ASME Research Publication, 4th Edition, 1937.
 Heat extracted by cooling water = $2.42 \times 10^4 \times 5.95 \text{ Btu/sec}$
3. "ASME Power Test Codes of 1940", Part 5, Chapter 4, Flow Measurement, July 1940. = $\frac{2.42}{.0149} = 399 \text{ Btu/lb air}$
4. McAdams, W. H.: "Heat Transmission", 2nd Edition, McGraw, Hill Book Company, 1942.
 Sensible enthalpy of gases (Btu/lb air) 102
 Final sensible enthalpy of gases (Btu/lb air) 399
 Temperature of exhaust gases at exhaust (°F) 1575
5. Taylor, C. F., and Taylor, E. S.: "The Internal Combustion Engine", International Textbook Company, Scranton, Pa., 1938.
 Fuel air ratio 120% of chemically correct
6. "Temperature - Its Measurement and Control in Science and Industry", Reinhold Publishing Corp., 1941.
 f - mass fraction of residual gases
 P_x - pressure at point "x" on indicator card
 V_x - cylinder volume at point "x" on indicator card.
 E_R - sensible internal energy (Btu/lb products)
 E_G - sensible enthalpy (Btu/lb products)
 $M_o = \frac{102/982}{\text{eps}/2} = \frac{.0149}{1592} \times 120 = .001121$
 Clearance volume = 4.333 cubic inches
 Mass of f = .044

REFERENCES

1. King, W. T.: "Measurement of High Temperatures in High Velocity Gas Streams", Transactions, AIME, Vol. 62, 1943, pp. 431-431.
2. "Fluid Meters, Their Theory and Application", Part I, ASME Research Publication, 4th Edition, 1937.
3. "ASME Power Test Codes of 1940", Part 2, Chapter 4, Flow Measurement, July 1940.
4. McGraw, W. H.: "Heat Transmission", 3rd Edition, McGraw-Hill Book Company, 1942.
5. Taylor, C. F., and Taylor, E. S.: "The Internal Combustion Engine", International Textbook Company, Scranton, Pa., 1938.
6. "Temperature - Its Measurement and Control in Science and Industry", Reinhold Publishing Corp., 1941.

SAMPLE CALCULATIONS

Heat Exchanger Method

Conditions: Compression ratio 8
 Fuel air ratio 120% of chemically correct

Mean temperature in exhaust surge tank ($^{\circ}\text{F}$) 303
 Mass rate of water flow (lbs/min) 3.43
 "From Chart" air " (lbs/sec) 0.0149
 Rise in cooling water temperature ($^{\circ}\text{F}$) 104

Heat extracted by cooling water = $\frac{3.43 \times 104}{60} = 5.95 \text{ Btu/sec}$

From Indicator diagram

$$V_4 = 42.664 \text{ in}^3 = \frac{5.95}{0.0149} = 399 \text{ Btu/lb air}$$

Sensible enthalpy of gases @ 303 $^{\circ}\text{F}$ (Btu/lb air) 102
 Rise in sensible enthalpy of gases (Btu/lb air) 399
 Final sensible enthalpy of gases (Btu/lb air) 501
 Temperature of exhaust gases at exhaust ($^{\circ}\text{F}$) 1555

Indicator Card Method

Conditions: Compression ratio 8
 Fuel air ratio 120% of chemically correct

Symbols: M_a = mass of air (lbs/cycle)

V_{cyl} = actual cylinder volume (cubic inches)

f = mass fraction of residual gases

MP_x = pressure at point "x" on indicator card

V_x = cylinder volume at point "x" on indicator card.

E_s = sensible internal energy (Btu/lb products)

H_s = sensible enthalpy (Btu/lb products)

$$M_a = \frac{\text{lbs/sec}}{\text{rps/2}} = \frac{0.0149}{1593} \times 120 = .001121$$

Clearance volume = 5.333 cubic inches

Assume $f = .044$

SAMPLE CALCULATIONS

Heat Exchanger Method

Conditions: Compression ratio 8
Fuel air ratio 120% of chemically correct

Mean temperature in exhaust gases (°F) 303
Mass rate of water flow (lbs/min) 3.43
" " " " (lbs/sec) 0.0149
Rise in cooling water temperature (°F) 104

$$\text{Heat extracted by cooling water} = \frac{3.43 \times 104}{60} = 5.95 \text{ Btu/sec}$$

$$= \frac{5.95}{0.0149} = 399 \text{ Btu/lb air}$$

Temperature of exhaust gases at exhaust (°F) 1222
Rise in sensible enthalpy of gases (Btu/lb air) 399
Sensible enthalpy of gases at 303°F (Btu/lb air) 103

Indicator Card Method

Conditions: Compression ratio 8
Fuel air ratio 120% of chemically correct

Symbols: M_a - mass of air (lbs/cycle)
 V_{cyl} - actual cylinder volume (cubic inches)

r - mass fraction of residual gases
 P_x - pressure at point "x" on indicator card
 V_x - cylinder volume at point "x" on indicator card

H_a - sensible internal energy (Btu/lb products)
 H_s - sensible enthalpy (Btu/lb products)

$$M_a = \frac{1 \text{ lb/sec}}{1593} \times 120 = 0.00121$$

Clearance volume = 2.333 cubic inches

Assume $r = 0.04$

From Indicator diagram

$$P_x = 65.95 \text{ #/in}^2$$

$$V_x = 37.73 \text{ in}^3$$

$$V_{\text{chart}} = \frac{37.73}{1728} \times \frac{.956}{.001121} = 18.6 \text{ ft}^3$$

$$V_{\text{clearance}} = \frac{5.333}{1728} \times \frac{.956}{.001121} = 2.63 \text{ ft}^3$$

From Chart

$$V_5 = 60 \text{ ft}^3 \quad \therefore r = \frac{V_{\text{clearance}}}{V_5} = \frac{2.63}{60} = .044$$

From Indicator diagram

$$V_4 = 42.664 \text{ in}^3$$

$$V_4 = \frac{42.664}{1728} \times \frac{.956}{.001121} = 21.05$$

From Chart

$$E_{s4} = 885 - 336 = 549$$

$$E_{s5} = 692 - 336 = 356$$

$$(1 + r)H_{se} = E_{s4} - rE_{s5} + \frac{P_a(V_1 - V_2)}{J}$$

$$= 549 - .044 \times 356 + \frac{14.7 \times 144 \times 18.32}{778}$$

$$= 583.6$$

$$H_{se} = 610 \text{ Btu/# air + fuel}$$

$$T_e = 2345^\circ R = 1885^\circ F$$

SLOAN LABORATORY
ENGINE GER. BORE 3.25" STROKE 4.5"

TABLE I

From Indicator diagram

$$P_x = 62.92 \frac{W}{in^2} \quad V_x = 37.73 in^3$$

$$V_{obart} = \frac{37.73}{1728} \times \frac{0.92}{0.00121} = 18.6 \text{ ft}^3$$

$$V_{olence} = \frac{2.332}{1728} \times \frac{0.92}{0.00121} = 2.63 \text{ ft}^3$$

From Chart

$$V_2 = 60 \text{ ft}^3 \quad \therefore r = \frac{V_2}{V_1} = \frac{60}{2.63} = 22.81$$

From Indicator diagram

$$V_4 = 42.664 in^3$$

$$V_4 = \frac{42.664}{1728} \times \frac{0.92}{0.00121} = 21.02$$

From Chart

$$E_4 = 882 - 336 = 546$$

$$E_2 = 692 - 336 = 356$$

$$(1 - r)H_{23} = E_4 - E_2 + \frac{20(V_1 - V_2)}{3}$$

$$249 = 546 - 356 + \frac{16.7 \times 144 \times 18.35}{320}$$

$$= 583.6$$

$$H_{23} = 610 \text{ Btu/lb air + fuel}$$

$$T_3 = 2342^\circ R = 1832^\circ F$$

SLOAN LABORATORY

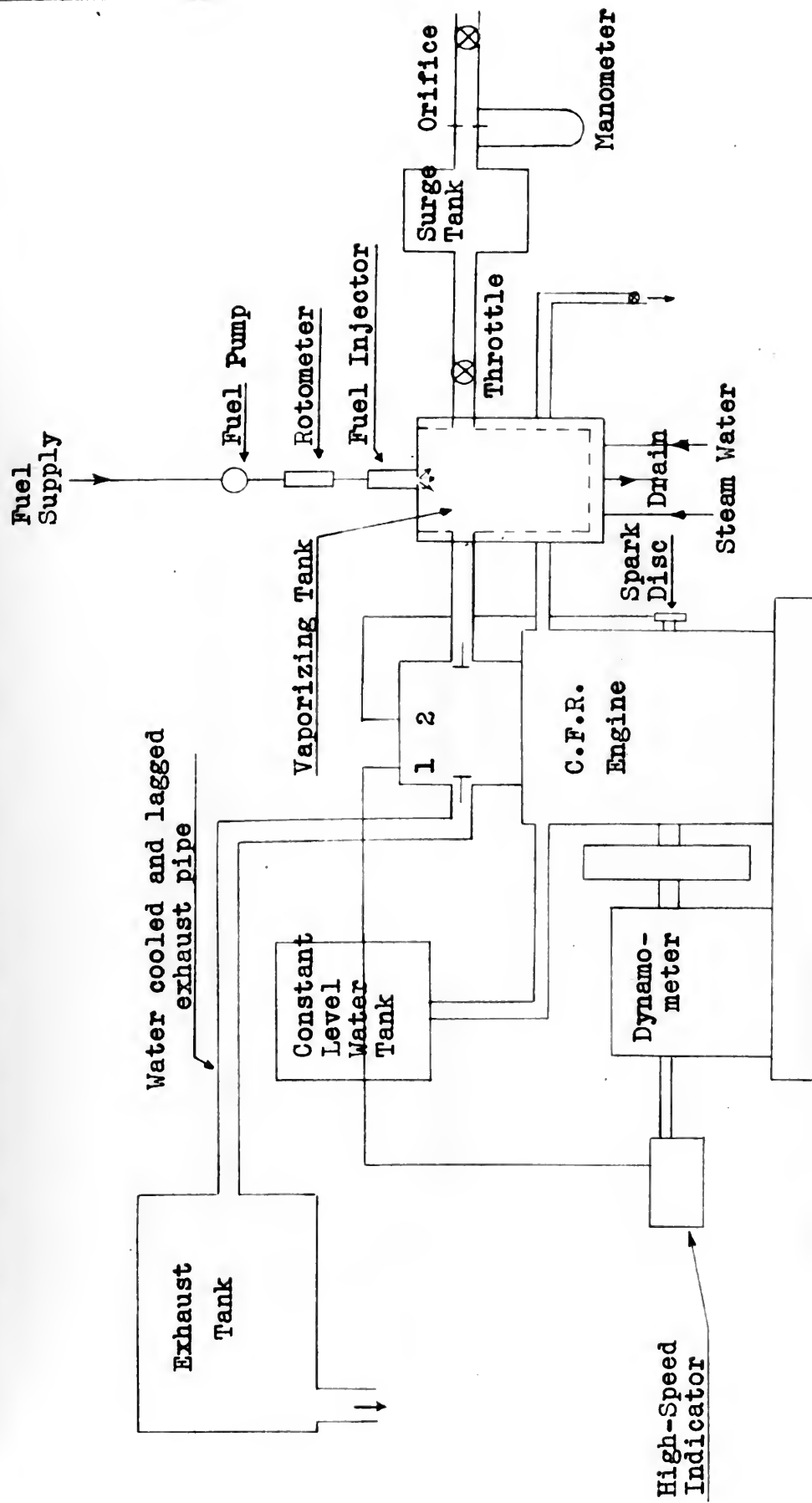
ENGINE C.E.R. BORE 3.25" STROKE 4.5"

REMARKS	DRIE TIME	RPM	B.L. "Hg	COMP. RATIO	OIL PRES.	TEMP OIL	TEMP		P ₁ "Hg	P _E "Hg	T ₁ °F	AIR CONVS. #/SEC	FUEL CONVS. #/SEC	F A	S.A. °	AIR P ₁ "Hg	ΔP "H ₂ O	FUEL ROTO 85°F	WATER		THERMO COUPLES				
							T ₁ °F	T ₂ °F											W #/MIN	I °F	2 °F	3 °F	G.E. °F		
	4/12 1600	1	1610	7.25	7	40	160	212	28.30	28.92	141	.0150	.0112	.0798	20	2892	12.30	9.90	54	170	2.35	330	325	337	1613
		2	1610	7.28	7	40	160	212	28.30	28.92	140	.0148	.0108	.0731	20	2892	12.00	9.34	54	163	3.33	304	302	314	1744
		3	1610	7.10	7	40	160	212	28.30	28.92	155	.0148	.0108	.0665	20	2892	12.00	8.85	56	174	3.38	328	323	338	1840
		4	1610	6.50	7	40	160	212	28.30	28.92	155	.0148	.0108	.0599	20	2892	12.00	8.37	56	170	3.56	330	330	340	1820
		5	1610	5.60	7	40	160	212	28.30	28.92	155	.0148	.0108	.0532	20	2892	12.00	7.85	57	173	3.51	338	332	348	1844
	4/24 1700	6	1593	6.95	8	38	136	212	28.11	28.73	150	.0149	.0109	.0798	20	2873	12.14	9.85	56	160	3.43	300	300	310	1626
		7	1580	6.85	8	38	136	212	28.11	28.73	150	.0149	.0109	.0731	20	2873	12.14	9.37	56	155	3.67	300	300	310	1705
		8	1587	6.60	8	38	134	212	28.11	28.73	150	.0149	.0109	.0665	20	2873	12.14	8.90	56	157	3.71	301	301	311	1712
		9	1590	6.25	8	39	132	212	28.11	28.73	150	.0154	.0114	.0599	20	2873	13.00	8.55	56	162	3.47	300	300	310	1706
		10	1590	5.40	8	39	129	212	28.11	28.73	150	.0156	.0113	.0532	20	2873	13.30	8.07	56	175	3.09	303	302	313	1733
	4/24 2200	11	1595	8.60	9	41	126	212	28.11	28.73	134	.0156	.0114	.0798	20	2873	13.30	10.08	55	162	2.90	279	278	288	1459
		12	1594	8.60	9	42	126	212	28.11	28.73	134	.0155	.0113	.0731	20	2873	13.30	9.58	55	166	2.92	286	286	297	1546
		13	1594	8.50	9	42	125	212	28.11	28.73	129	.0156	.0113	.0665	20	2873	13.30	9.14	55	176	2.80	300	299	310	1605
		14	1587	8.10	9	42	125	212	28.11	28.73	129	.0156	.0113	.0599	20	2873	13.30	8.61	55	160	3.20	300	300	310	1565
		15	1577	8.10	9	42	125	212	28.11	28.73	129	.0156	.0113	.0532	20	2873	13.30	8.06	54	158	3.25	299	299	309	1577

TABLE I

Date		Description		Amount	
1900	Jan 1	Balance		100.00	
1900	Jan 15	Received from A. B.		50.00	
1900	Feb 1	Received from C. D.		25.00	
1900	Mar 1	Received from E. F.		75.00	
1900	Apr 1	Received from G. H.		100.00	
1900	May 1	Received from I. J.		150.00	
1900	Jun 1	Received from K. L.		200.00	
1900	Jul 1	Received from M. N.		250.00	
1900	Aug 1	Received from O. P.		300.00	
1900	Sep 1	Received from Q. R.		350.00	
1900	Oct 1	Received from S. T.		400.00	
1900	Nov 1	Received from U. V.		450.00	
1900	Dec 1	Received from W. X.		500.00	
1900	Dec 31	Total		2500.00	

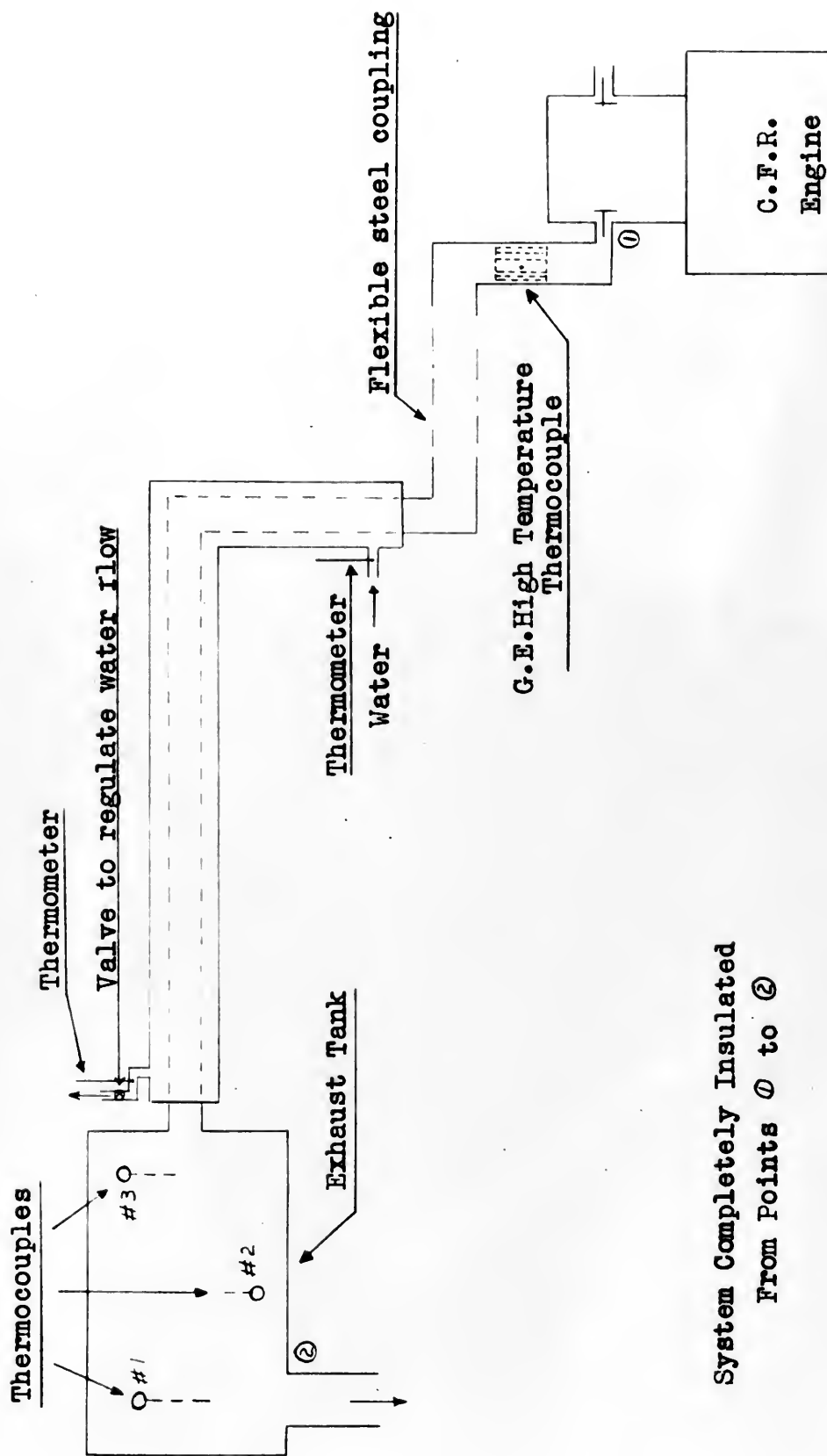
Received of the Treasurer of the
Board of Directors of the
City of New York



1. Indicator Cylinder Attachment
2. Spark Plug

SLOAN LABORATORY
EXPERIMENTAL C.F.R. ENGINE SETUP

Fig. 1



System Completely Insulated
From Points ① to ②

EXHAUST SYSTEM SETUP

Fig. 2



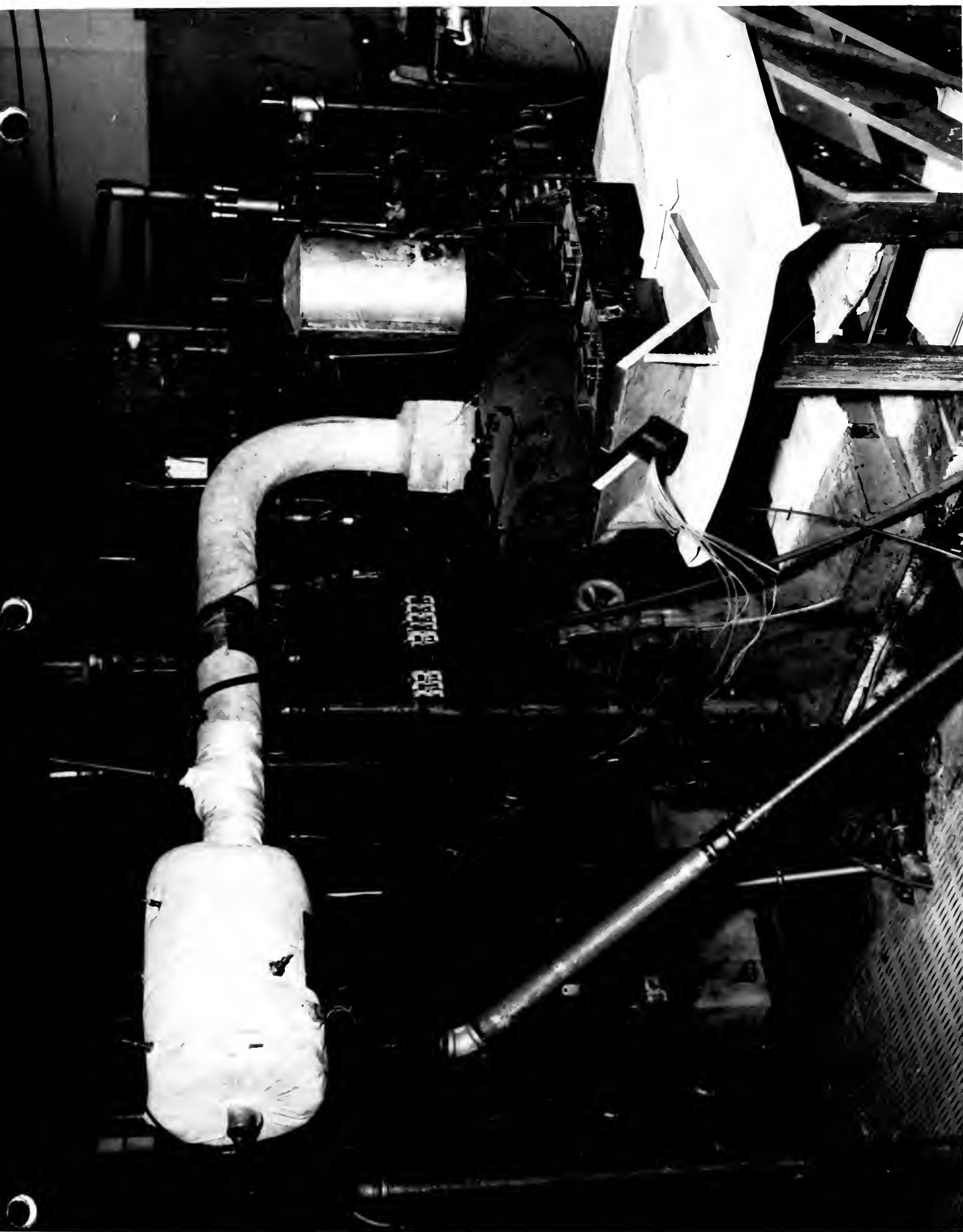
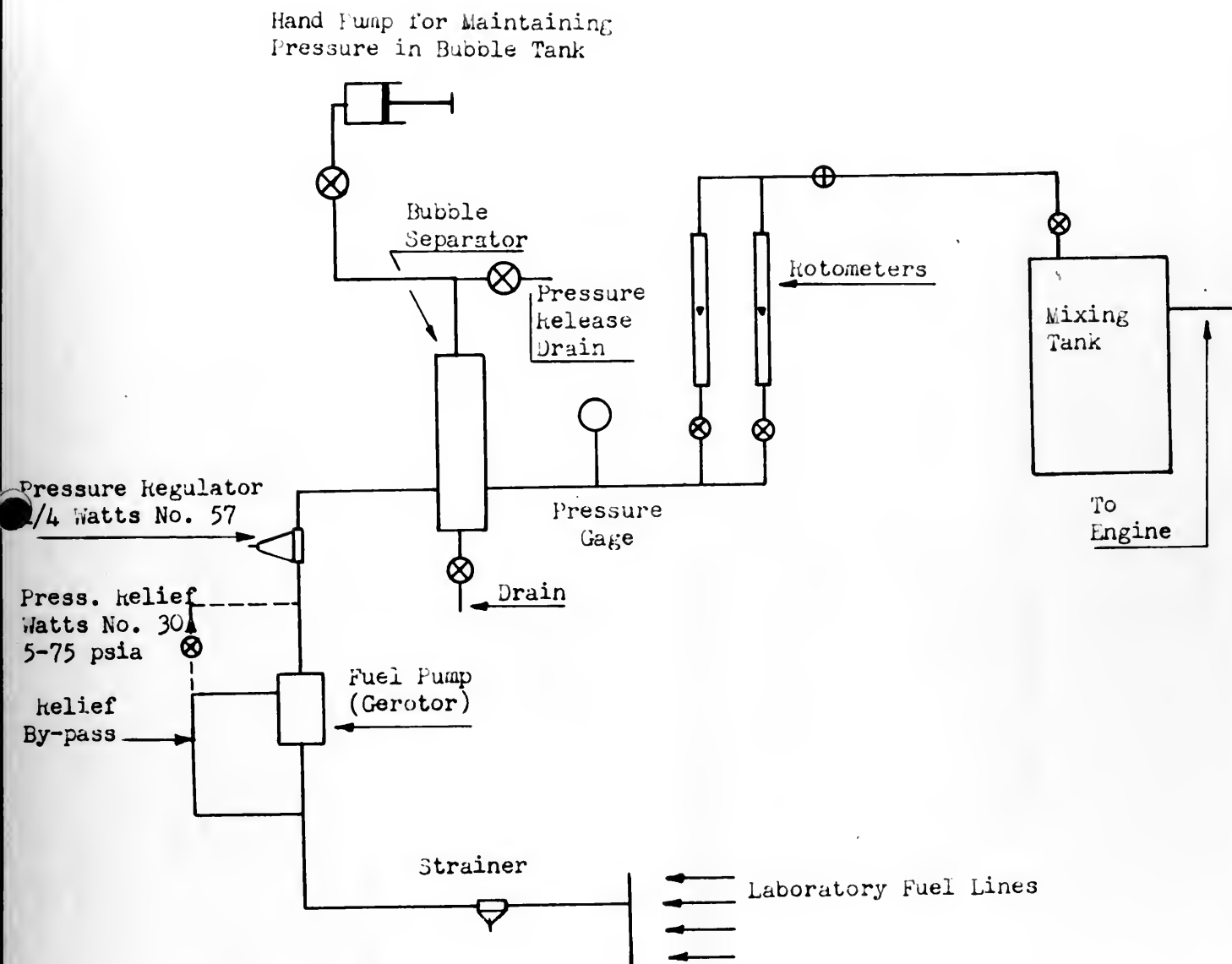


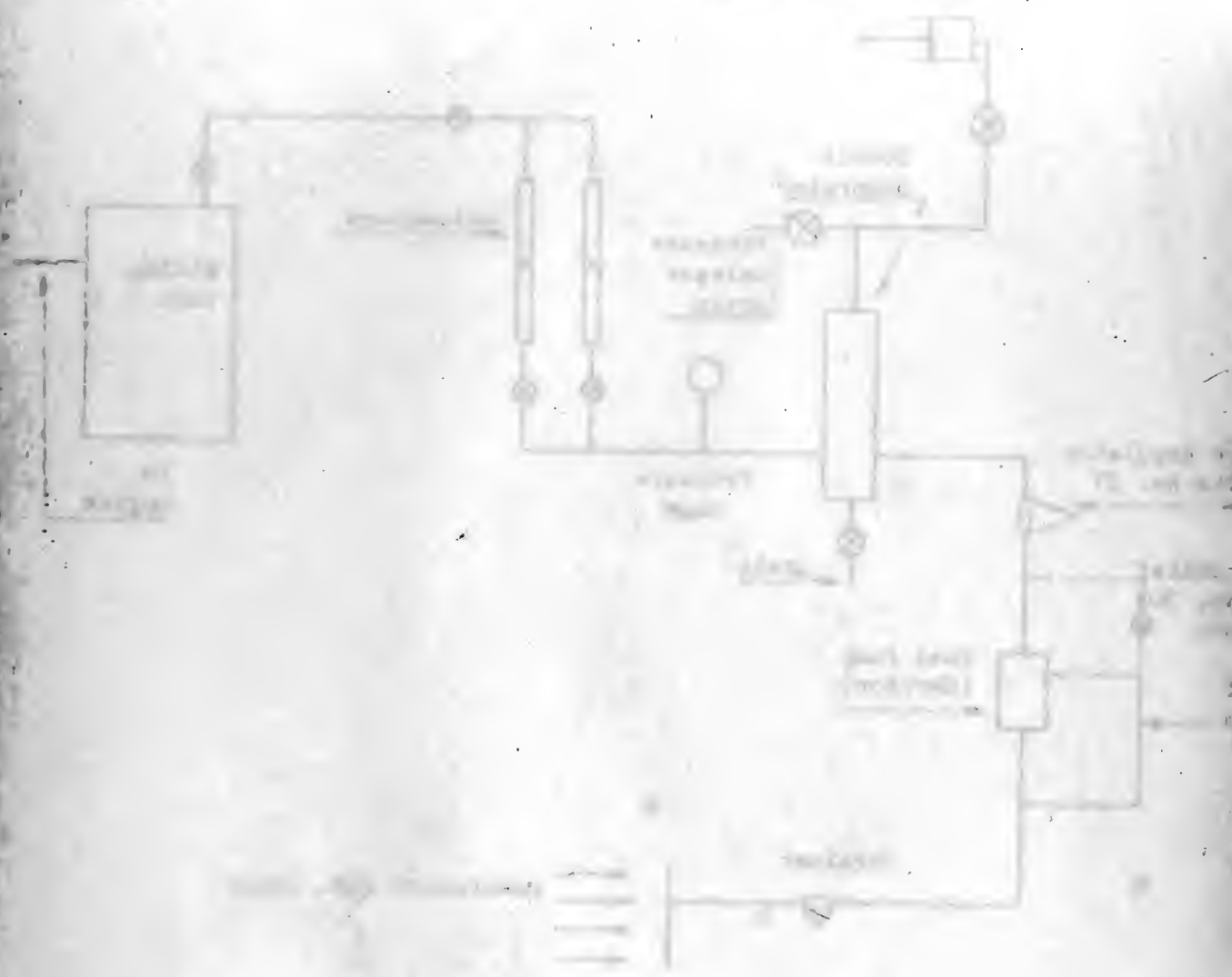
Fig 3

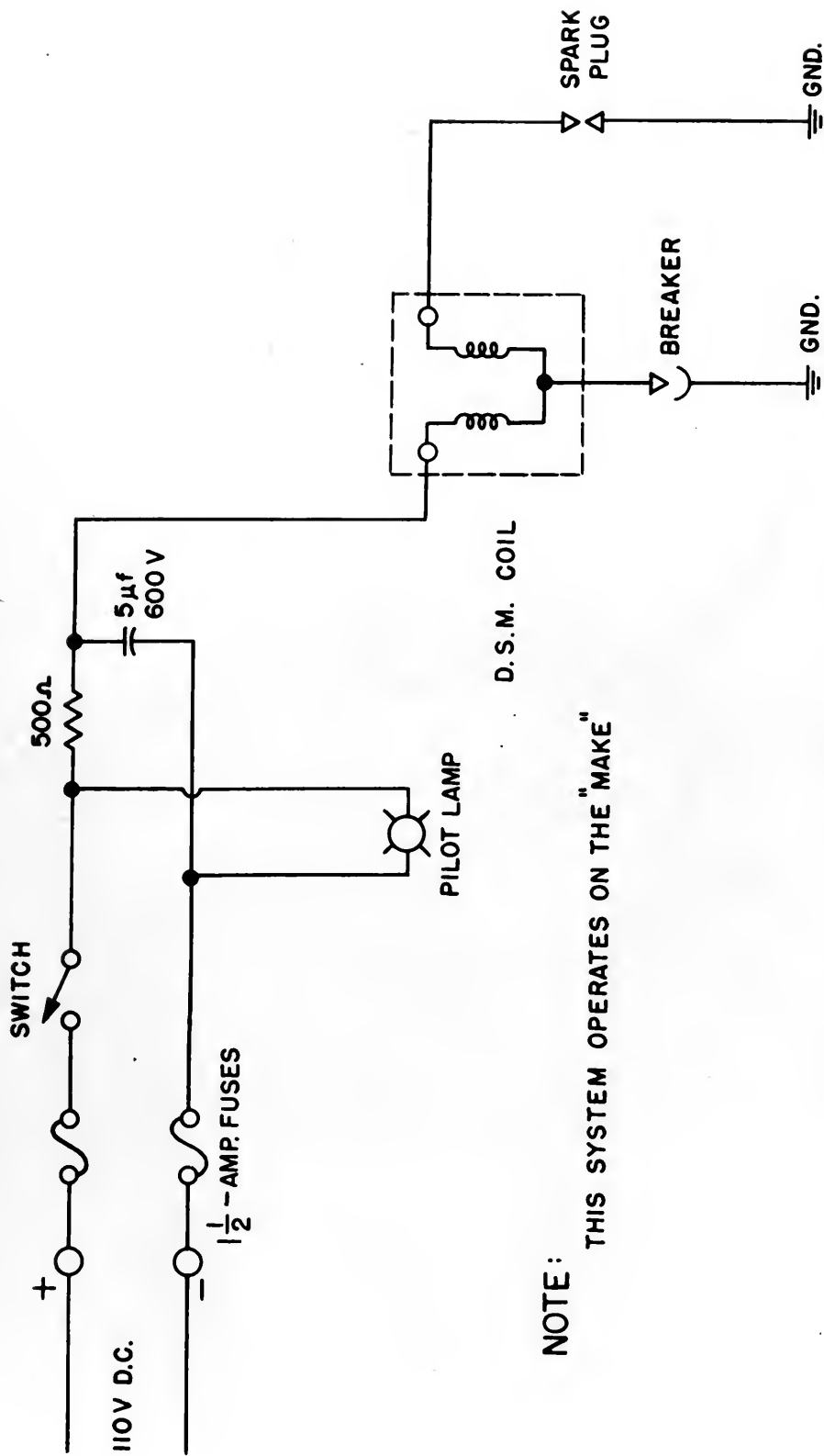




SCHEMATIC DIAGRAM OF FUEL SYSTEM

Fig.4

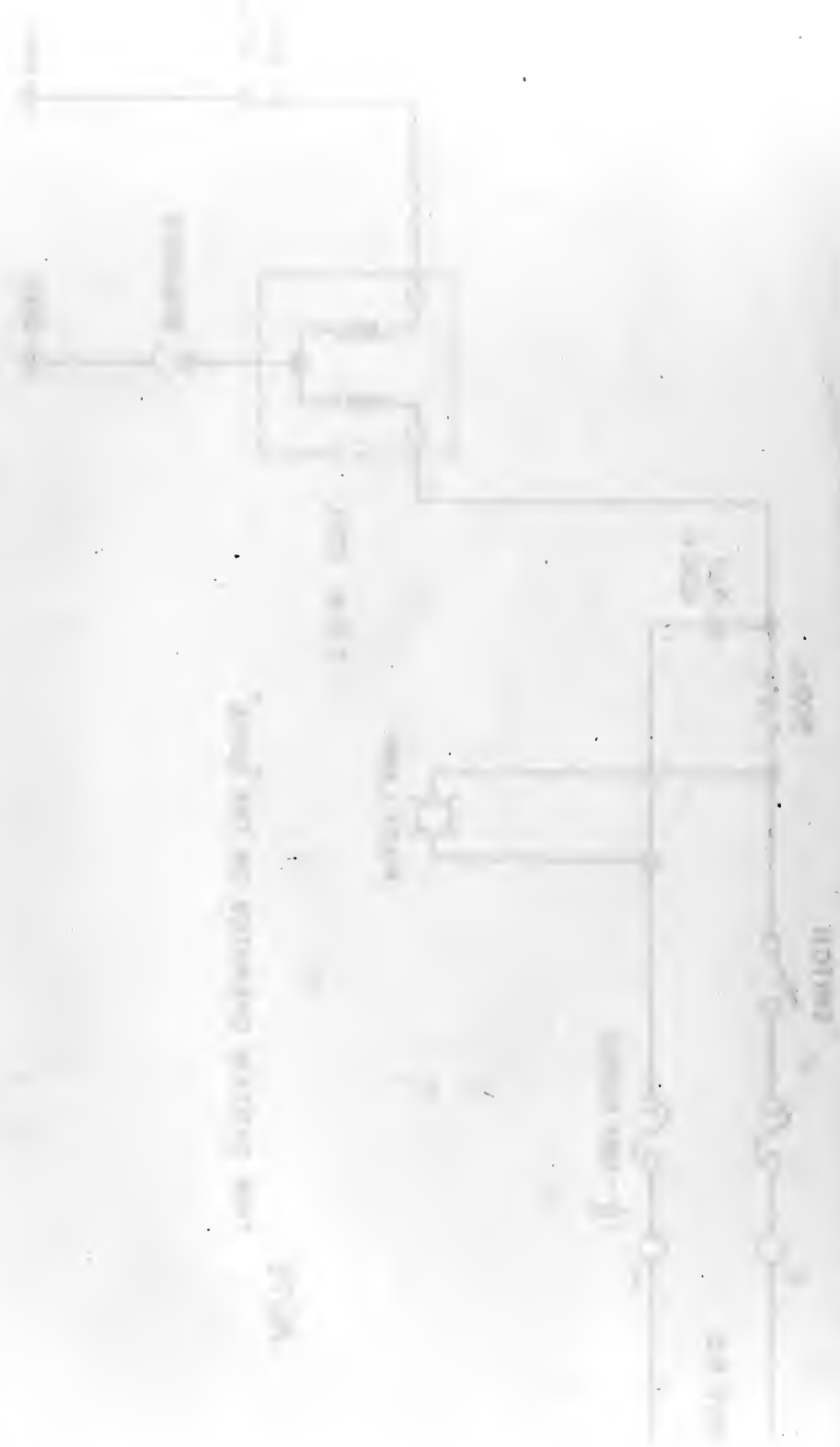




NOTE: THIS SYSTEM OPERATES ON THE "MAKE"

WIRING DIAGRAM OF IGNITION SYSTEM

WIRING DIAGRAM OF AUTOMATIC FEEDER



110V AC

10A

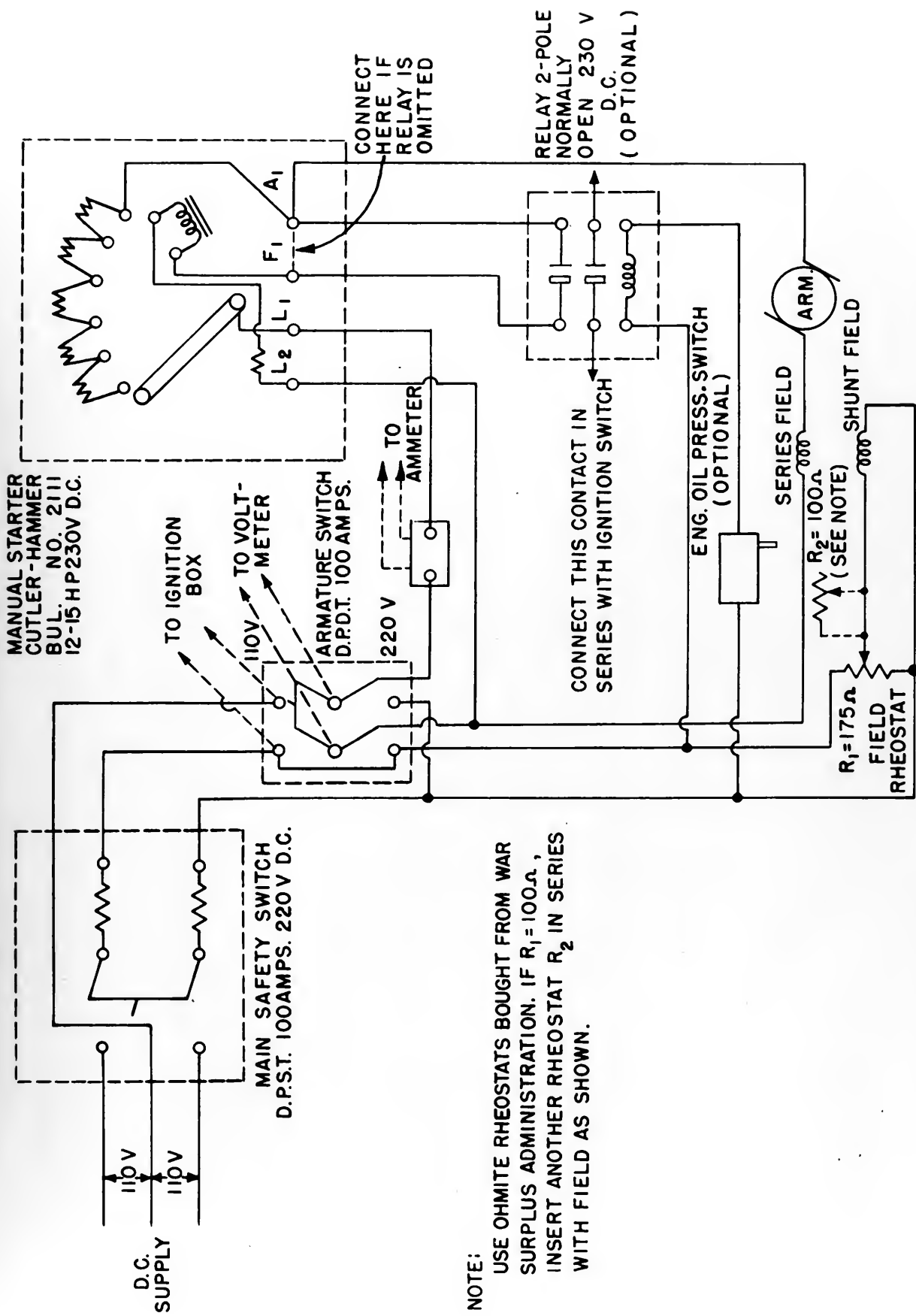
100W

100V

SWITCH



Fig. 6



WIRING DIAGRAM OF STAR DYNAMOMETER - CFR ENGINE SETUP

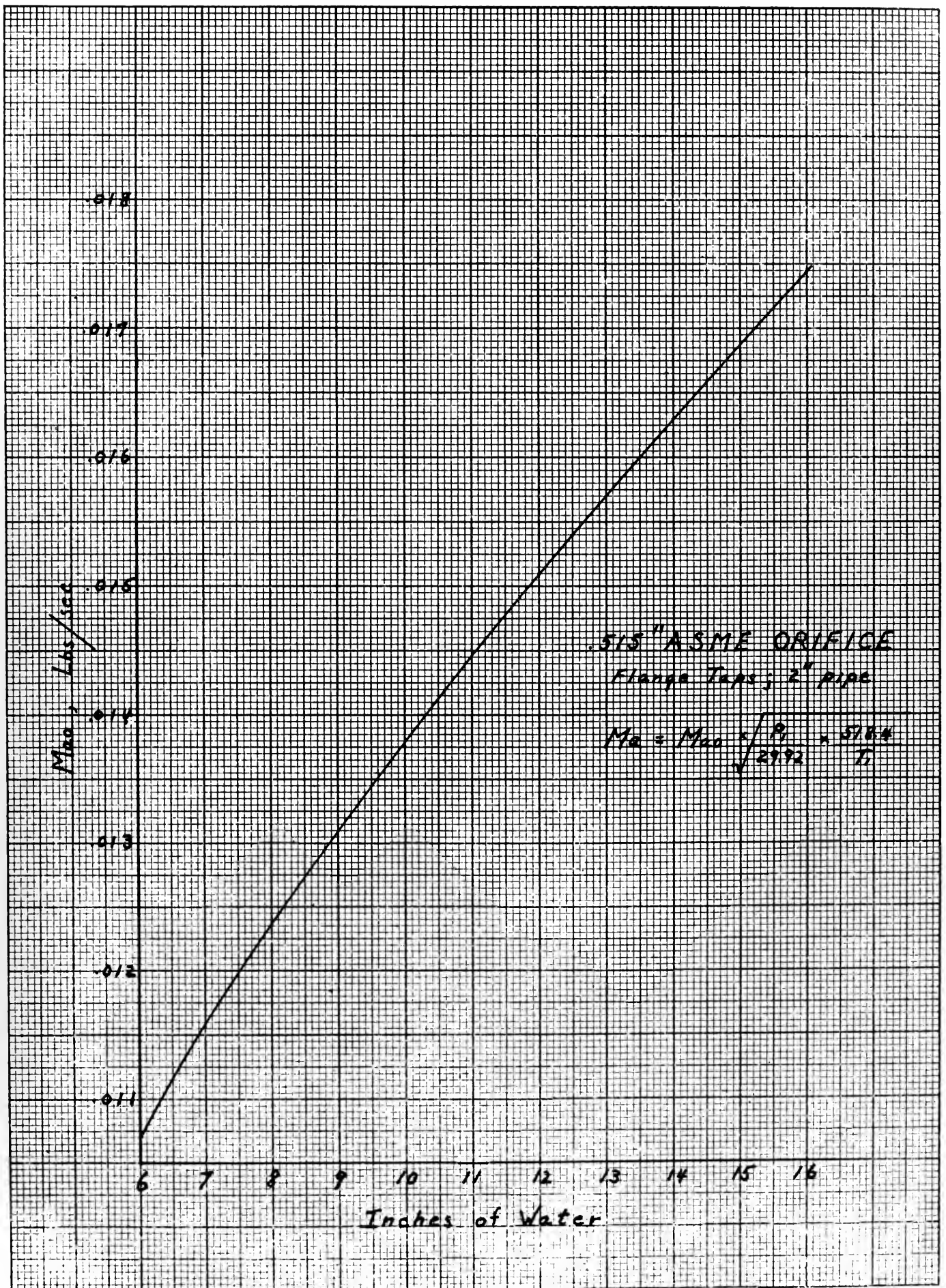
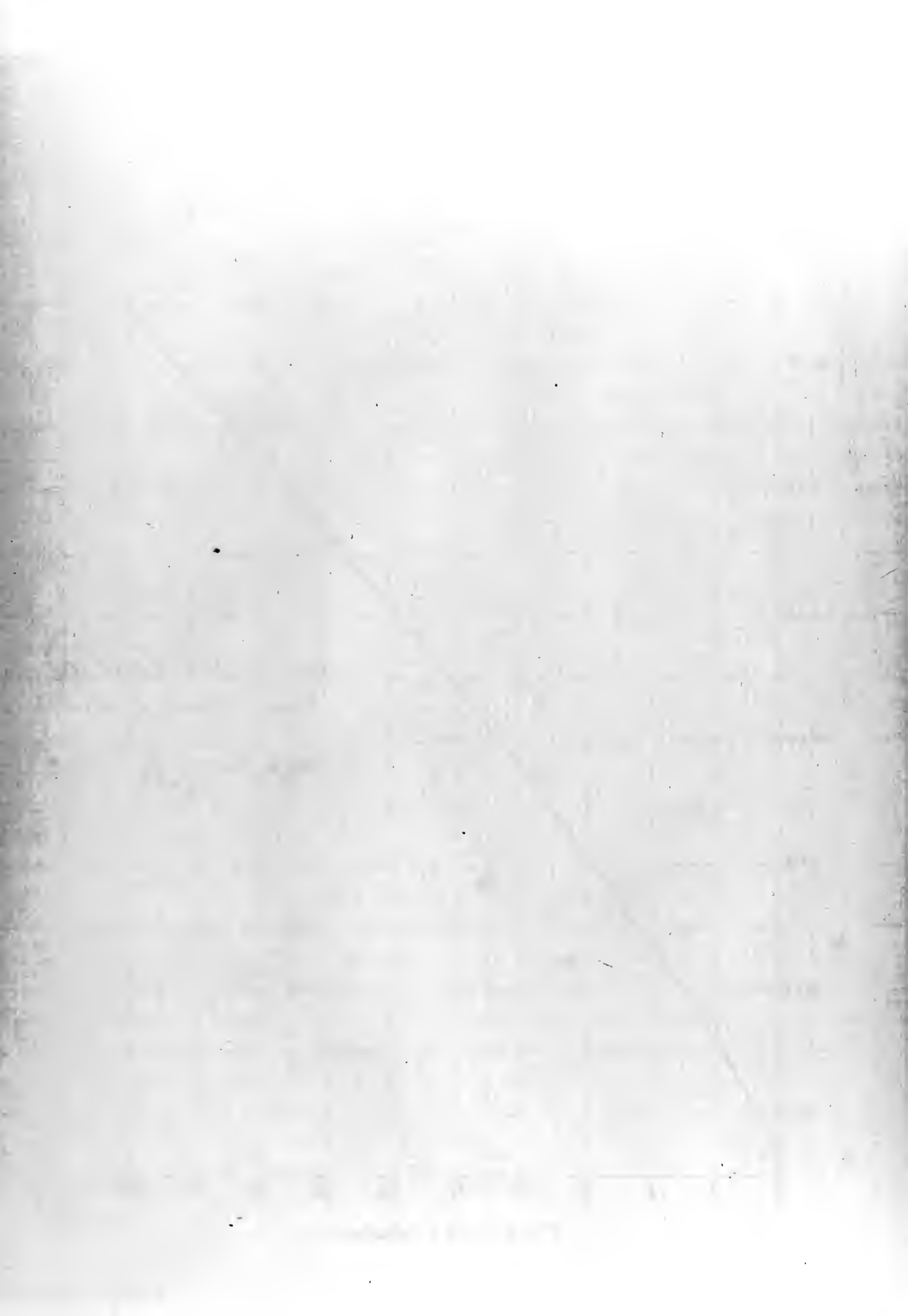


FIG. 8



Fuel Calibration Curves

Rotometer # H8-2990

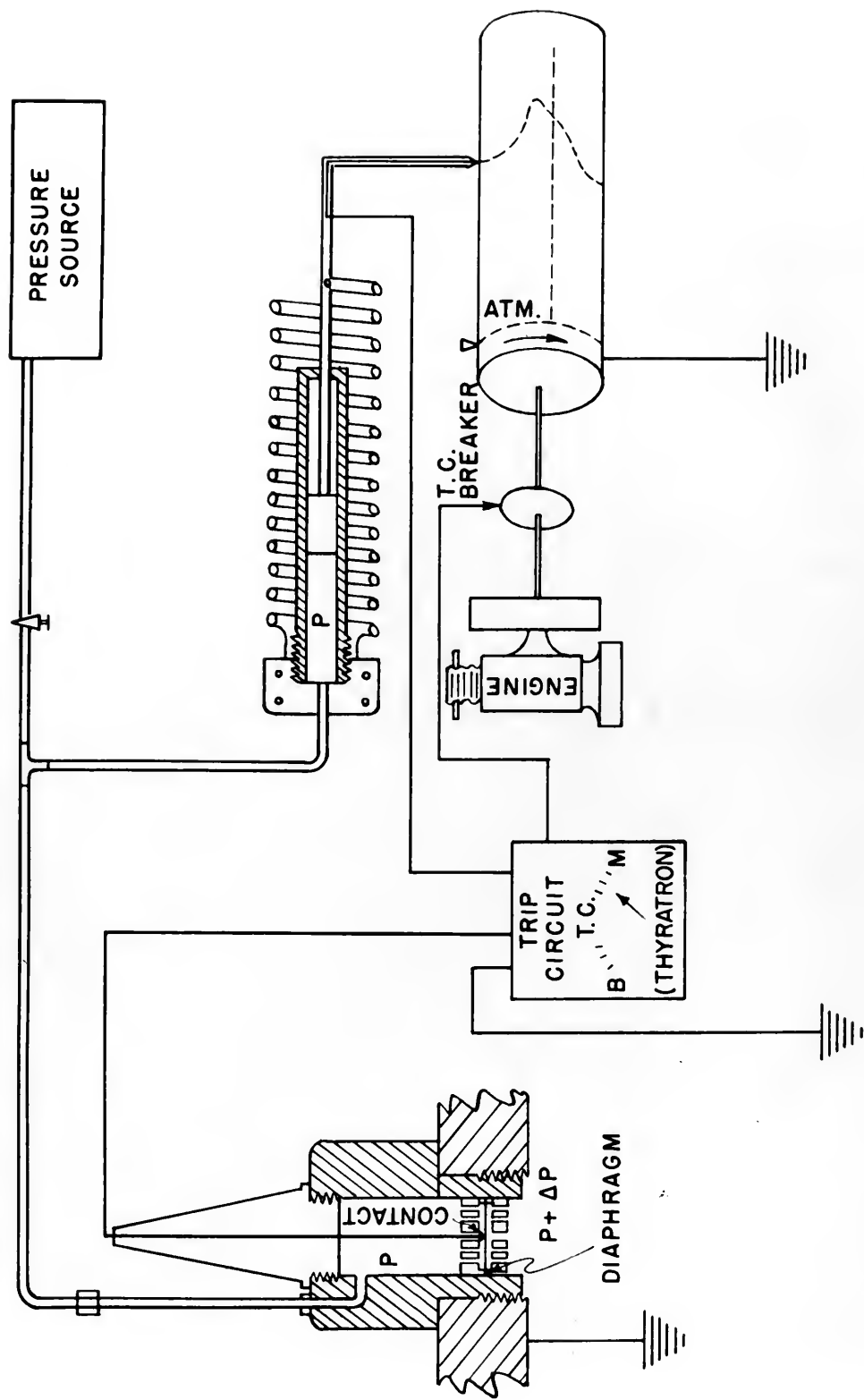
Fuel Flow, lbs/sec

85°F

75°F

Rotometer Reading

FIG. 9

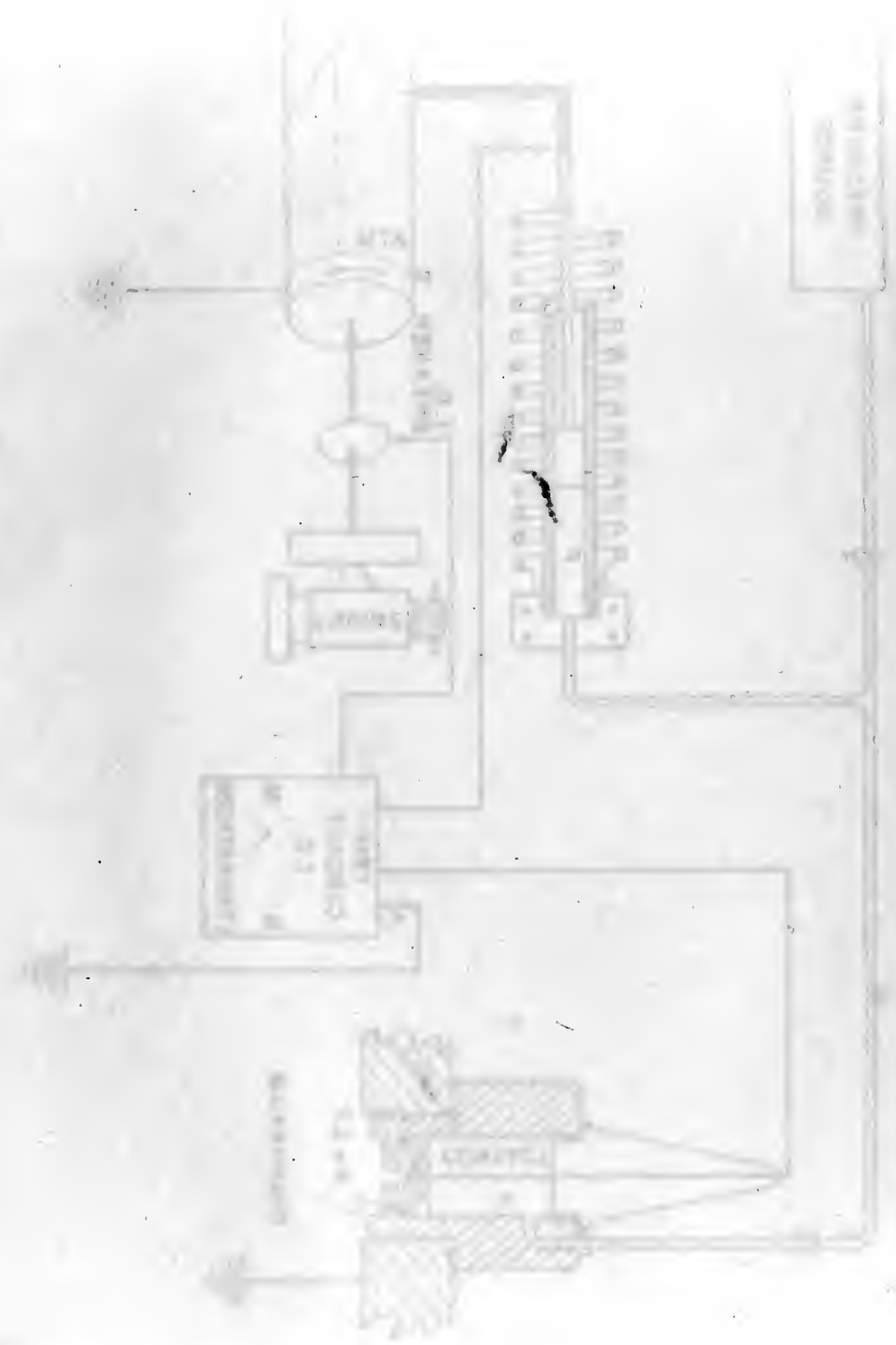


HIGH SPEED INDICATOR

M. I. T.

010557

WATER TREATMENT PLANT



E = INTERNAL ENERGY, B.T.U. ABOVE CO., HO (VAPOR), O., AND AIR-N. AT 100F
 Q_u = INT ENERGY OF COMBUSTION, AT 100F OF UNBURNED FUEL IN THE EQUILIBRIUM MIXTURE AT T. WHEN T = 2880°R. Q_u = 336.
 E = E - Q_u
 H = ENTHALPY, E + J(PV)
 H = E + J(PV) [H = H + Q_u

P = PRESSURE, LBS/SQ IN (DASHED LINES)
 V = VOLUME, CU FT (SOLID LINES)
 S = ENTROPY, ABOVE CO., HO (VAPOR), O., AND AIR-N. EACH AT ONE ATMOSPHERE 100F
 T = TEMPERATURE, °R = (°F + 460)
 FUEL = (CH₄).

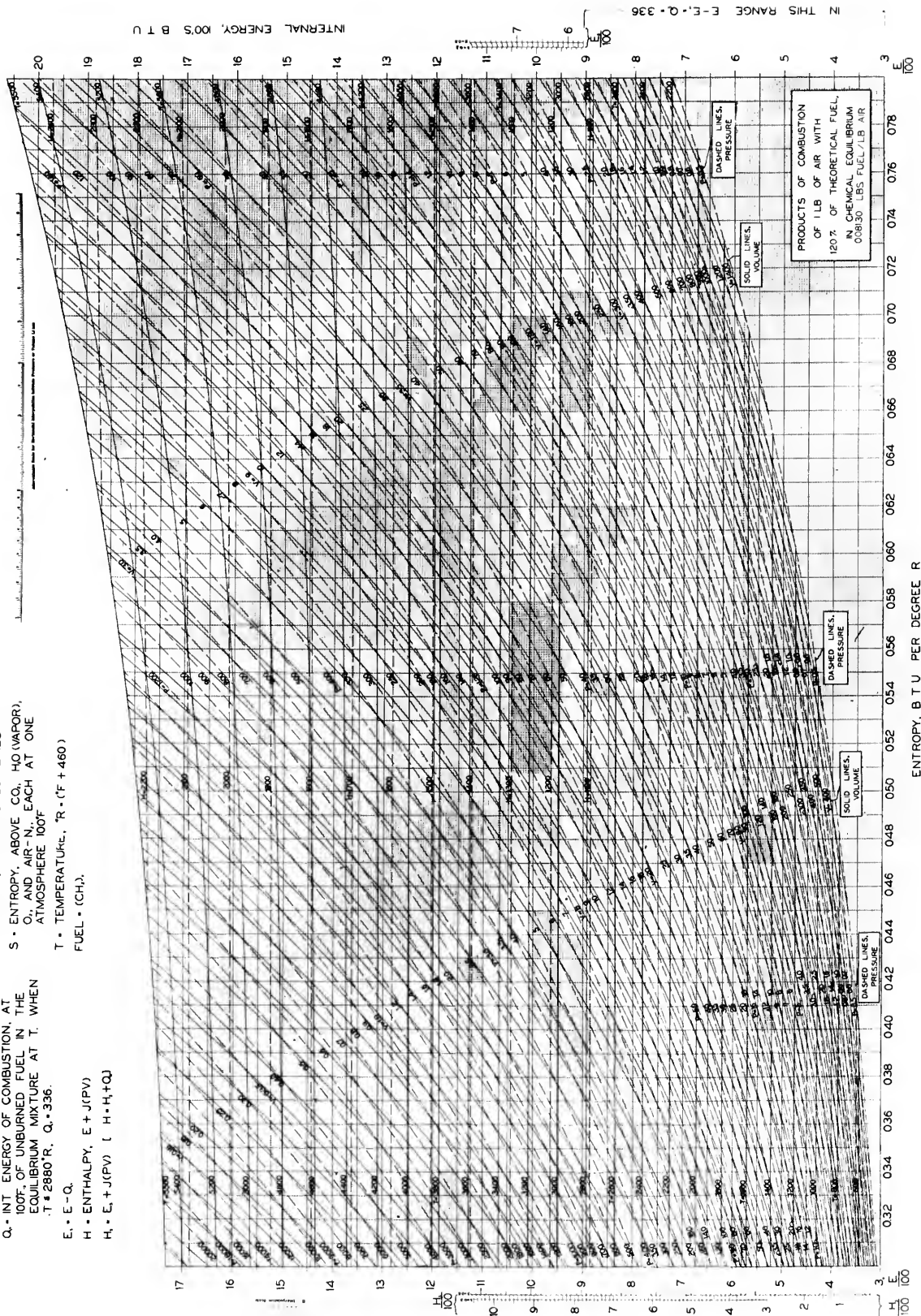


Fig. 11

EXHAUST GAS TEMPERATURE
VS
% THEORETICAL FUEL
(COMPRESSION RATIO = 7)

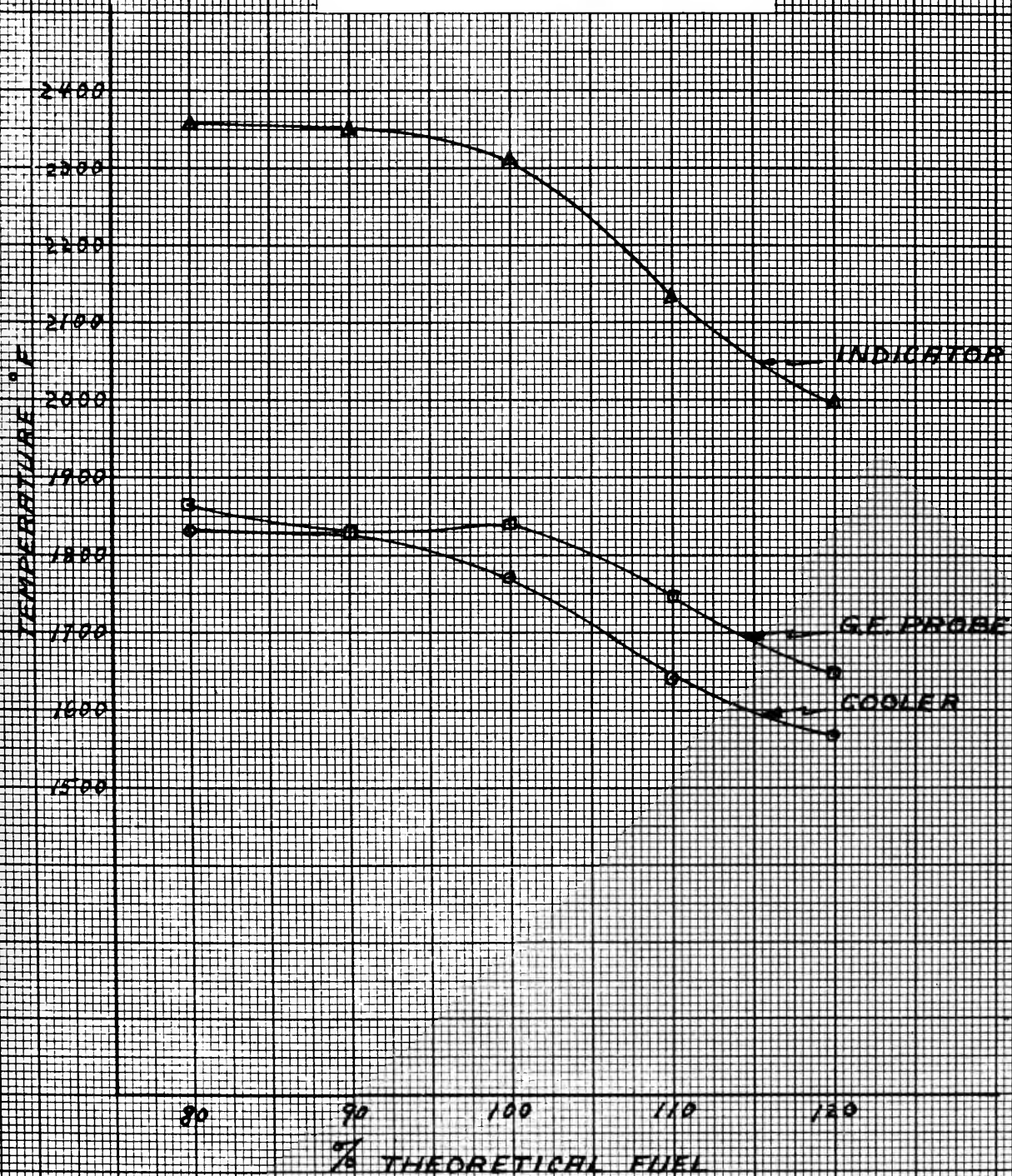


FIG. 12



EXHAUST GAS TEMPERATURE
VS
% THEORETICAL FUEL
(COMPRESSION RATIO = 8)

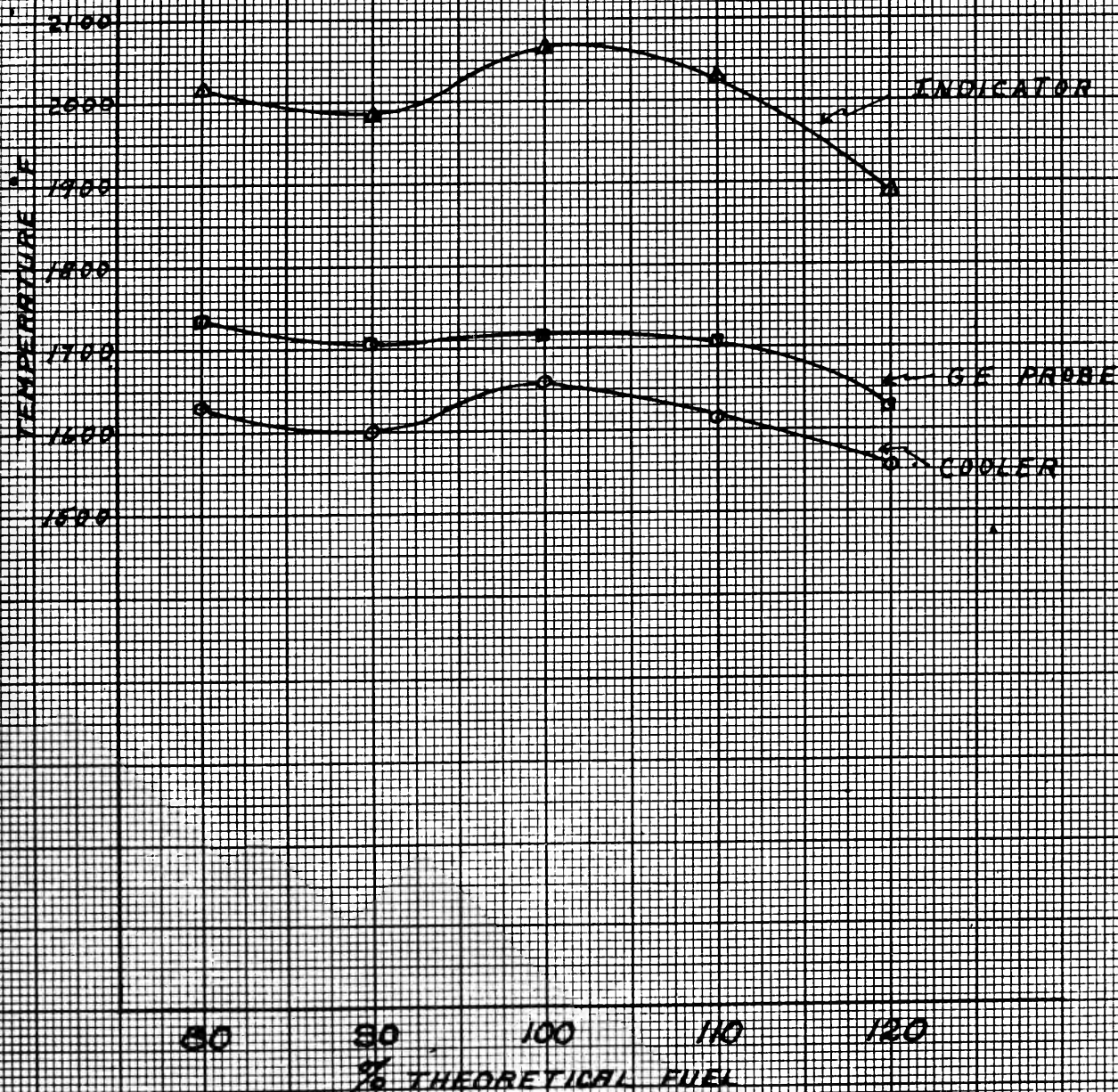


FIG. 13

EXHAUST GAS TEMPERATURE
VS
% THEORETICAL FUEL
(COMPRESSION RATIO = 9)

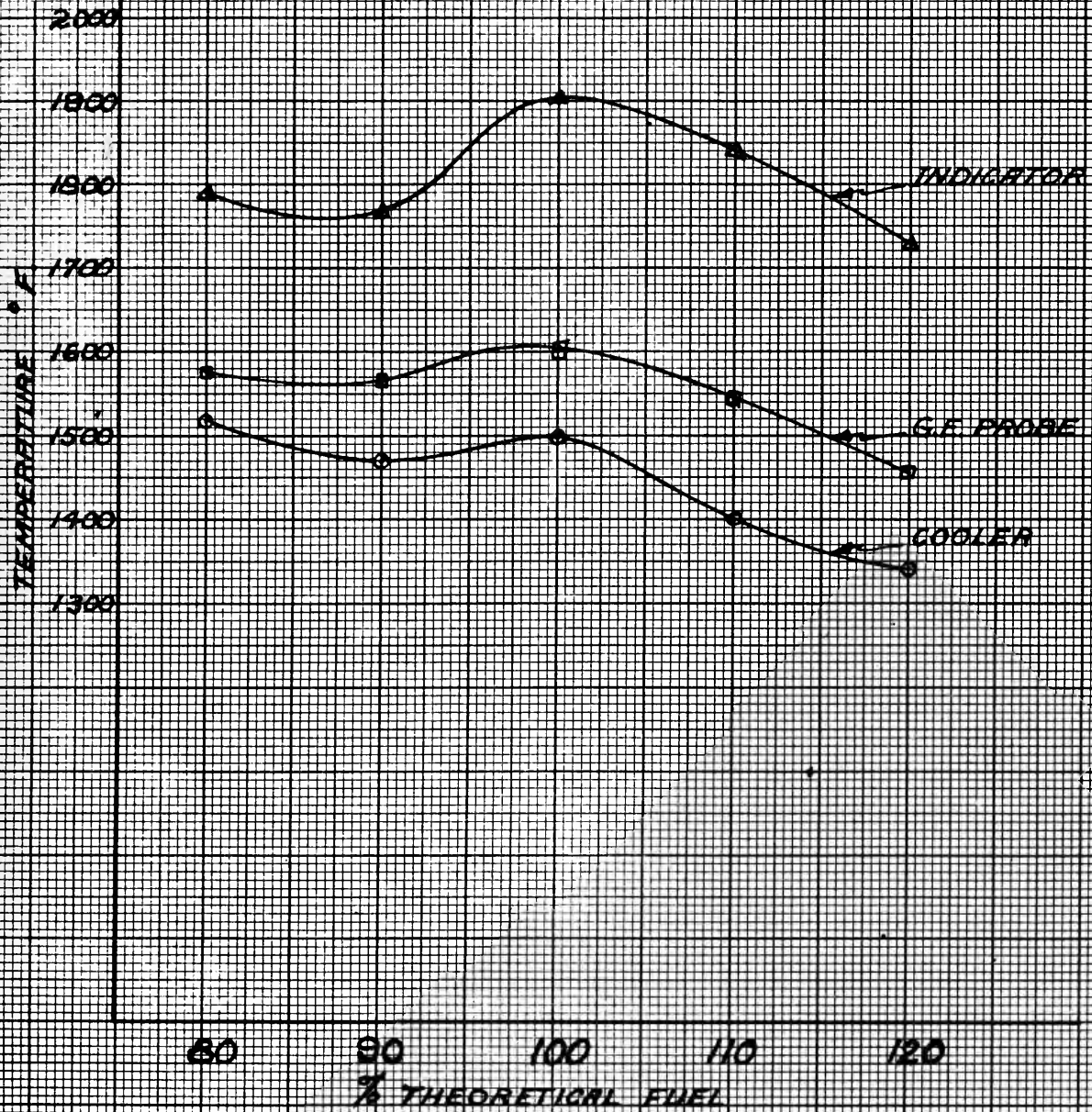


Fig. 14

EXHAUST GAS TEMPERATURE
VS
% THEORETICAL FUEL
(G.E. PROBE)

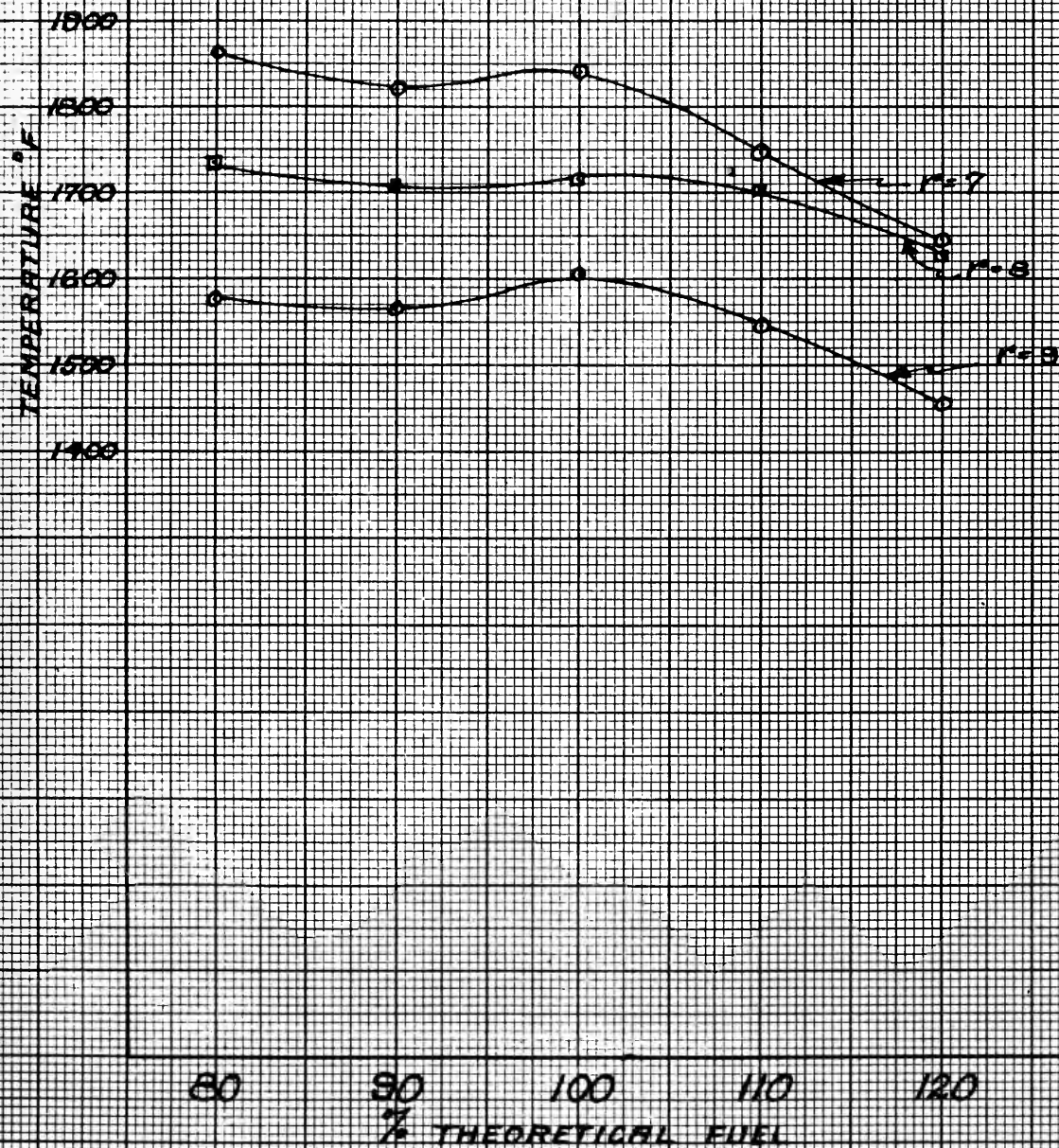


FIG. 15



EXHAUST GAS TEMPERATURE
VS
% THEORETICAL FUEL
(COOLER METHOD)

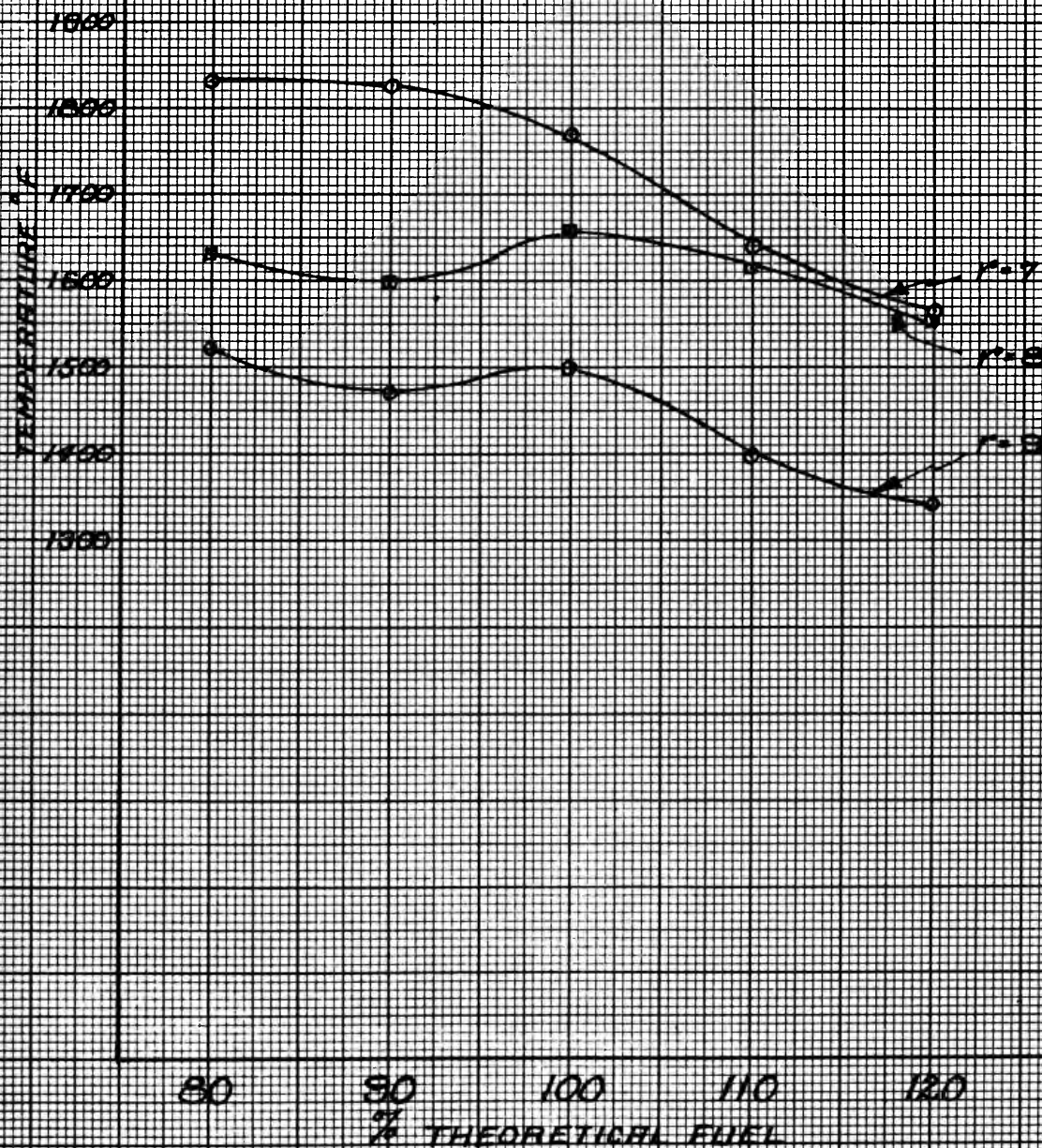
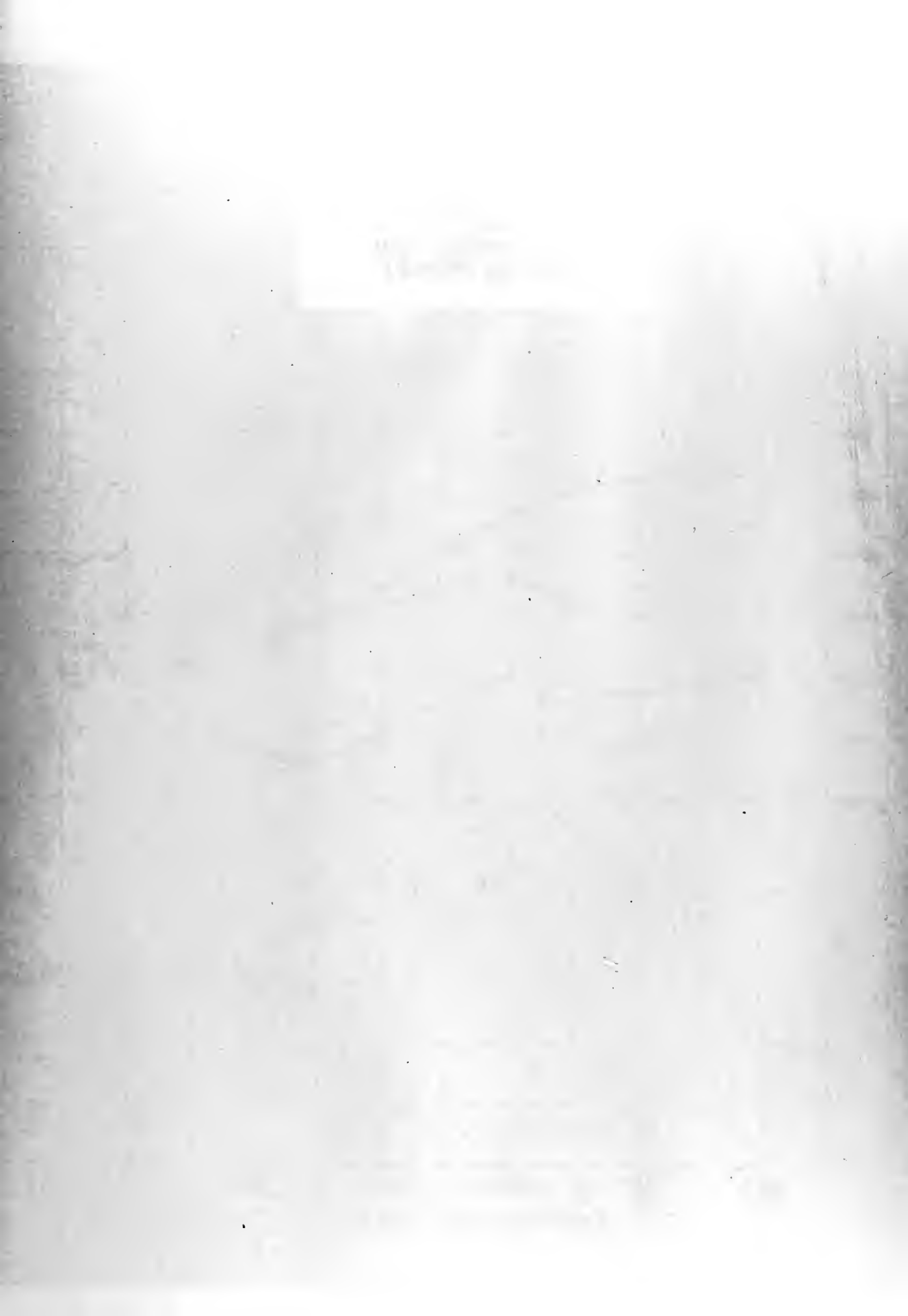


FIG. 16



EXHAUST GAS TEMPERATURE
VS
% THEORETICAL FUEL
(INDICATOR METHOD)



Fig. 17

[illegible]

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